

Physics 114 - May 2, 2006

- EXAM 3 STATUS



- Final EXAM

May 10

7:15 PM

↳ 10:15 PM

Lower Strong

Oh no!
It's my
Last physics
class.



- Calculator ok, **1** non-topologically-enhanced
8.5x 11 inch sheet of
formulas / Examples

- Q+A session Time/Loc. TBA

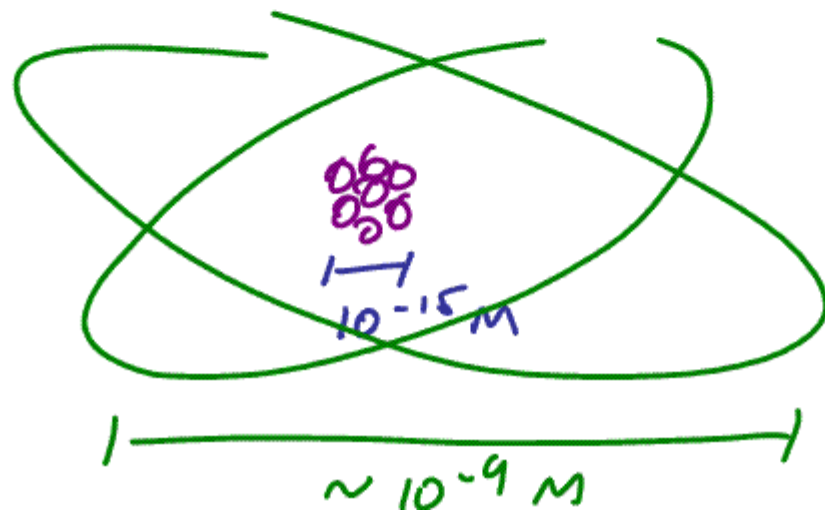
- Cumulative, like 2 term exams, 30-40% new
material

- Problem Set 14 ☺
+ Solns Posted on web

- Project Grades

- GRADE info endgame

Last Time - Nuclear Physics



Strong Nuclear force

↳ whatever that is ...

Nuclear density CONSTANT and
Very, Very Big

$$10^{17} \text{ kg/m}^3$$

Fermions
fill w/ \uparrow p, n
⇒ Nuclear Shell
Structure

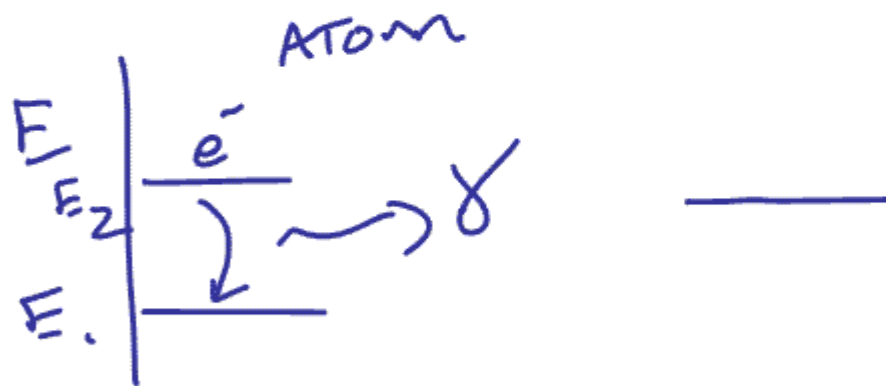
$$\Delta E \sim \text{MeV}$$



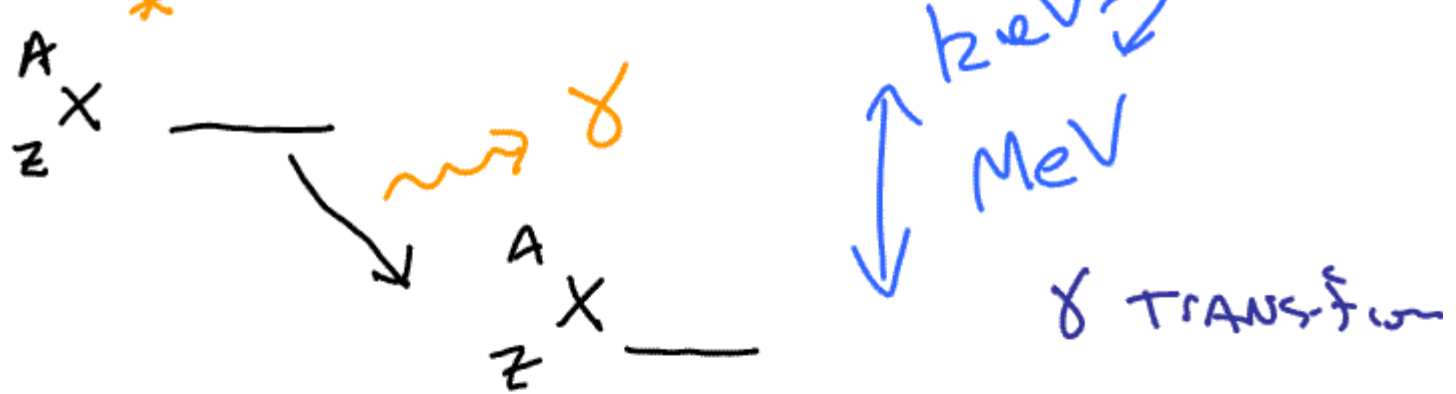
Get discrete energy levels
similar to ATOM

P, n exist in allowed states

n energy
 l orb. Ang. momentum
 m z comp
 s spin Ang. Mom



$A \equiv$ Atomic Mass,
 $Z =$ Atomic # \rightarrow # protons
 $N \equiv A - Z =$ # neutrons
 Z electrons



$$E = mc^2$$

MASS units $\frac{\text{MeV}}{c^2}$ Energy unit

$$M_{\text{electron}} = 0.511 \frac{\text{MeV}}{c^2}$$

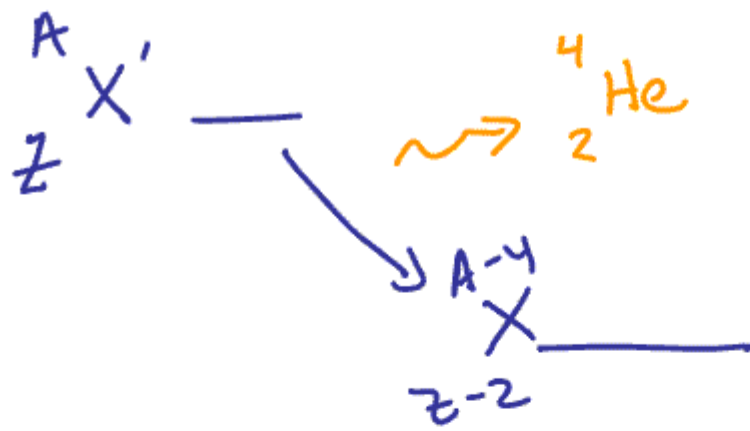
$$M_p \approx 938 \frac{\text{MeV}}{c^2}$$

1000 MeV

 \equiv GeV



β -RAY emission



α -particle

α radiation



SAME Z , different A \equiv isotopes
 # of neutrons

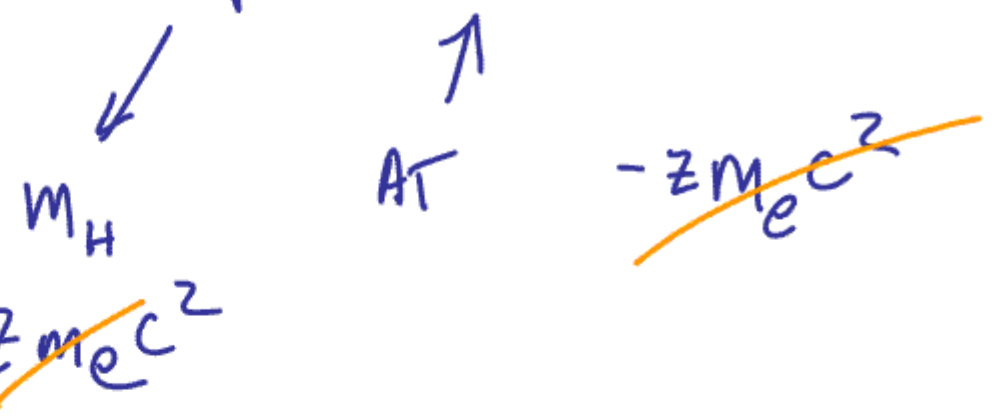
Binding Energy in nucleus

BE in H ATOM

$$m_e c^2 + m_p c^2 = m_H c^2 + 13.6 \text{ eV}$$

$$(A-Z)m_n c^2 + Z m_p c^2 = M_x c^2 + \text{TOT Nuc. B.E.}$$

CAN use
ATOMIC masses
+ ignore
 m_e
+ BE ok



$^{56}\text{Fe} \rightarrow 52103.5 \text{ MeV}/c^2$ $Z=26$
 $A-Z = 30$

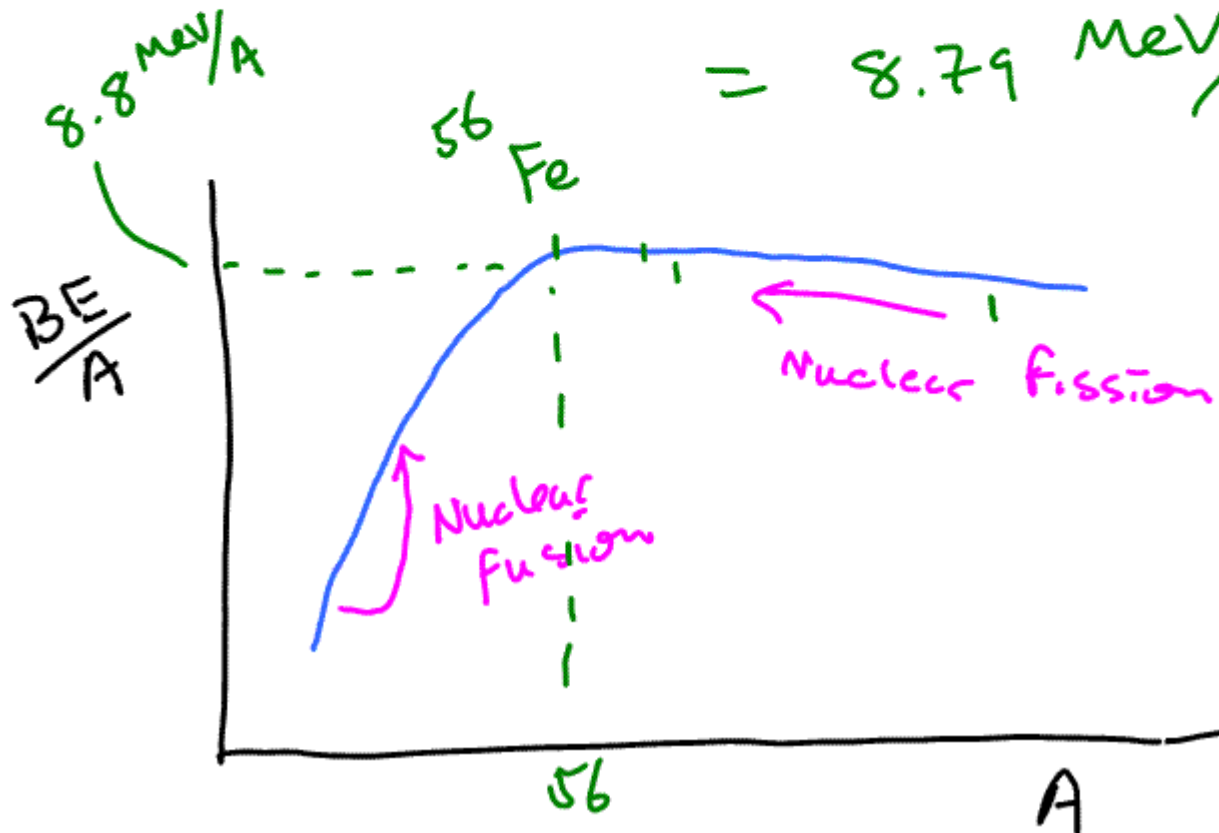
$$(BE)_{^{56}\text{Fe}} \left[(30)(939.57) + (26)(938.791) - 52103.5 \right]$$

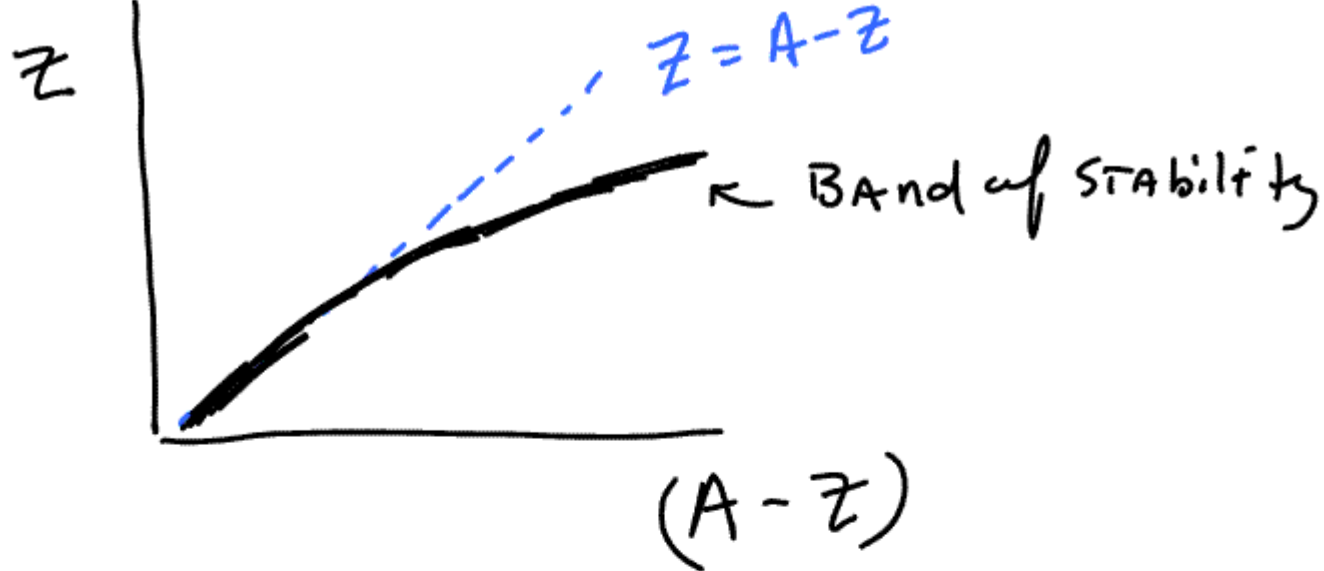
$$M_H = 938.791 \text{ MeV}/c^2 \quad m_n = 939.57 \text{ MeV}/c^2$$

$$(BE)_{56\text{Fe}} = 492.2 \text{ MeV}$$

$$\text{Binding Energy / Nucleon} = \frac{492.2 \text{ MeV}}{56}$$

$$= 8.79 \text{ MeV/nucleon}$$





naturally radioactive

SAMPLE N nuclei type X

$$\frac{dN}{dt} \propto N$$

$$\left(\frac{dN}{dt} = -\lambda N \right)$$

$$\frac{dN}{dt} \equiv A \equiv \text{Activity}$$

decay CONSTANT
 units $\frac{1}{\text{Time}} = \text{s}^{-1}$

$$\frac{dN}{dt} = -\lambda N$$

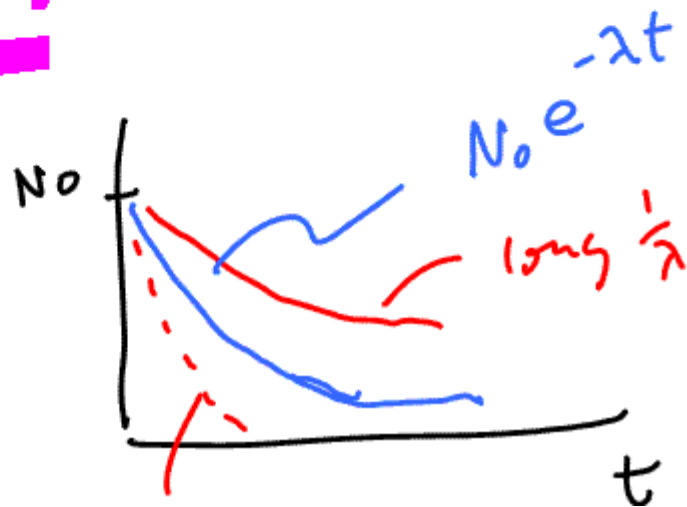
$$\frac{dN}{N} = -\lambda dt$$

$$\ln N = -\lambda t + c$$

$$N = N_0 e^{-\lambda t}$$

\downarrow
 $N(t)$

Exponential law
of
radioactive decay



$$\lambda N = \lambda N_0 e^{-\lambda t}$$

↑ Activity at time t ↑ mit Activity Activity # decays/sec

Meas. in Curies

$$1 \text{ Curie} = 3.7 \times 10^{10} \text{ dec./s}$$

mCi or μCi

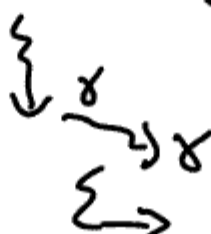
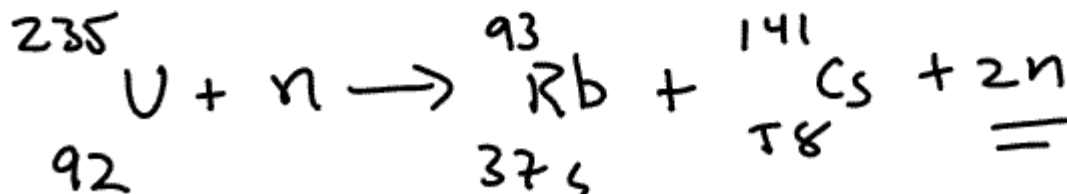
$T_{1/2}$ // Half life as time it takes for Activity to drop by factor of 2

$$A = A_0 e^{-\lambda t}$$

$$\frac{1}{2} A_0 = A_0 e^{-\lambda T_{1/2}}$$

$$T_{1/2} = \frac{1}{\lambda} \ln 2 = \frac{0.693}{\lambda}$$

Nuclear Fission



200 MeV
in
energy



neutrons released

START NEW fusion rxns

${}^{235}\text{U} \sim 0.7\%$

${}^{238}\text{U} \sim 99.3\%$

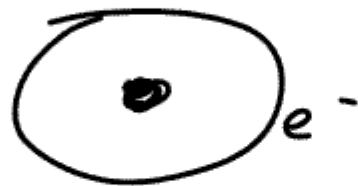
$\ll 1$ NO rxns

$= 1$ Critical

$\gg 1$ BOOM

Charged Particles

↳ interact w/ matter through ionization



ionizes Atoms
loses energy

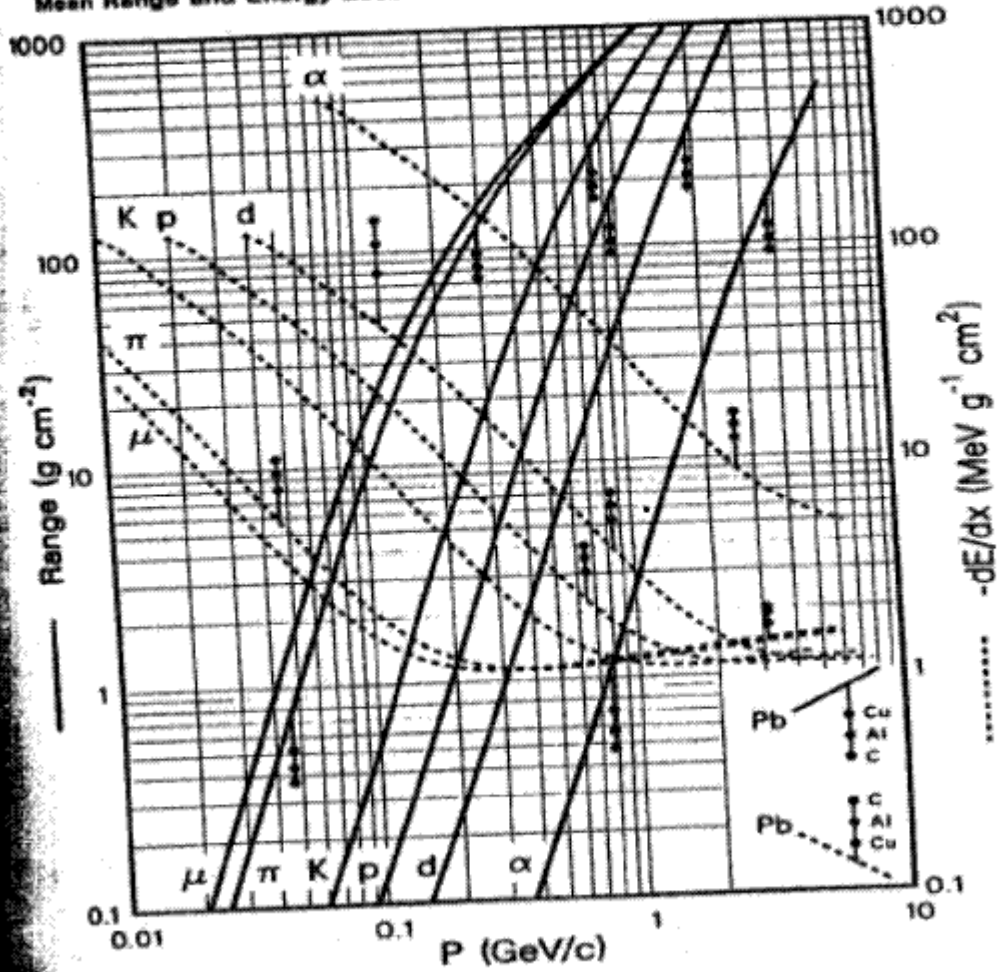
charged particle →

range and energy loss per cm depend
on the mass and momentum
of particle

Figure 2.3 Mean range and energy loss of charged particles in solids. Calculations use the Bethe-Bloch equation with density effect corrections. Refer to the cited reference for a discussion of assumptions and qualifications. (Particle Data Group, Rev. Mod. Phys. 56: S1, 1984.)

PARTICLE DETECTORS, ABSORBERS, AND RANGES

Mean Range and Energy Loss in Lead, Copper, Aluminum, and Carbon



get range in cm when mult by density

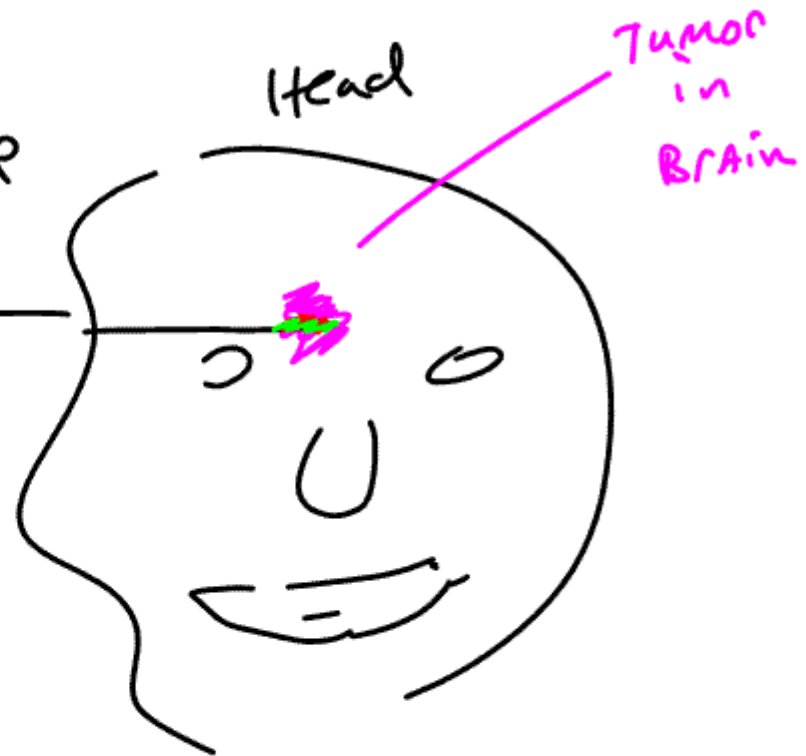
Energy loss in material per cm in STRANGE units

Mult. by density to get MeV/cm

CAN use this

Proton beam of momentum P

Protons
⇒
⇒



CAN TUNE momentum
So protons range out at
tumor location + kill
Tissue at that location
primarily

RANGE of protons of momentum P

ionization energy loss greatest when proton
almost stops ... Energy deposition and
Tissue destruction greatest at end of path