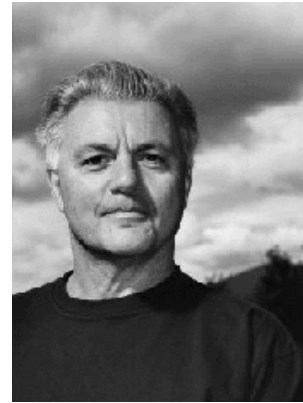


**Exit Welcome**  
**Enter Electromagnetism**

# The World According to Physics 121



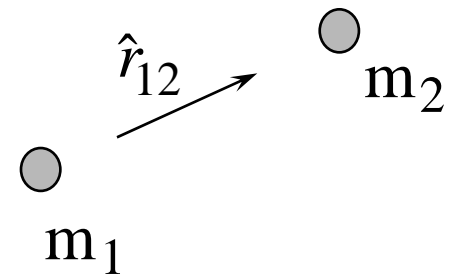
(Apologies to  
John Irving)

- Things, interactions and trajectories  $\vec{F} = \frac{d\vec{p}}{dt} = m\vec{a}$ 
  - Specified by geometry, mass and forces

- Forces

- Gravity:

$$\vec{F}_{12} = -G \frac{m_1 m_2}{r_{12}^2} \hat{r}_{12}$$



- Others: Tension, Normal, Friction

- Space and Time

- Euclidean with Galilean Invariance

- “ordinary” 3D space; “slow” velocities

# Moving Beyond 121...



- Things -- Bodies and Fields  
(e.g., gravitational)
  - Specified by geometry and mass and charge

- Forces

- Gravity:

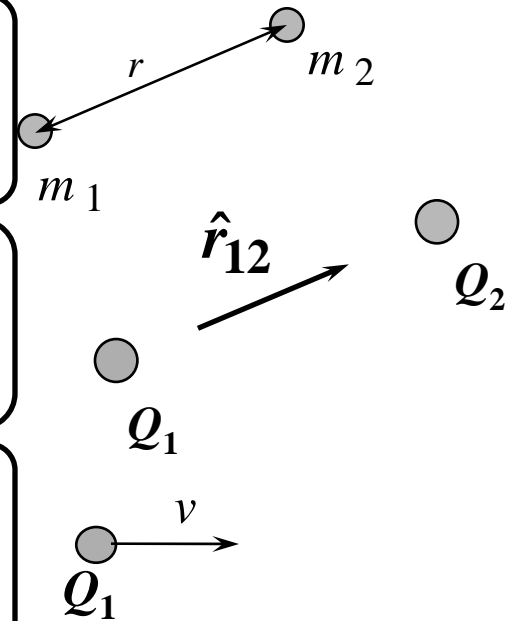
$$\vec{F}_{12} = -G \frac{m_1 m_2}{r_{12}^2} \hat{r}_{12}$$

- Electric:

$$\vec{F}_{12} = k \frac{Q_1 Q_2}{r_{12}^2} \hat{r}_{12}$$

- Magnetic:

$$\vec{F}_{1mag} = Q_1 \vec{v} \times \vec{B}$$



*Hmm... gravitational and electric forces look awfully similar...*

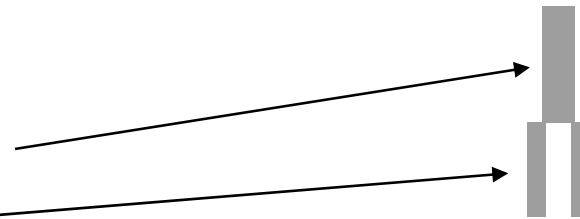
# Static Charges & Electroscopes

- **The Phenomena**
  - **Charge electroscope with rubber rod which has been rubbed with fur. Gold leaves separate.**
    - » **Bring same rubber rod close to top of electroscope. observe leaves separate further.**
    - » **Bring lucite rod (rubbed with silk) close to top of electroscope. observe leaves approach each other.**

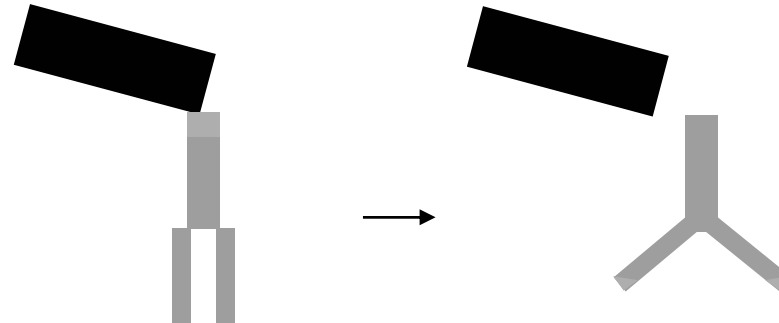
# More about charging electroscopes

- The electroscope is made out of conductors

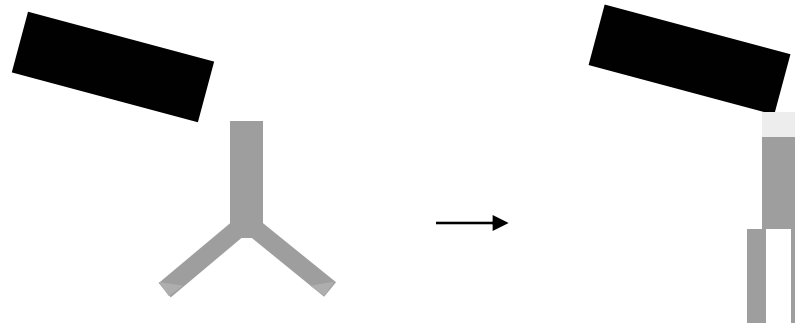
- conducting main electrode
- 2 conducting gold leaves



- Add some charge (rubber)



- Then add a different (?) charge (lucite)

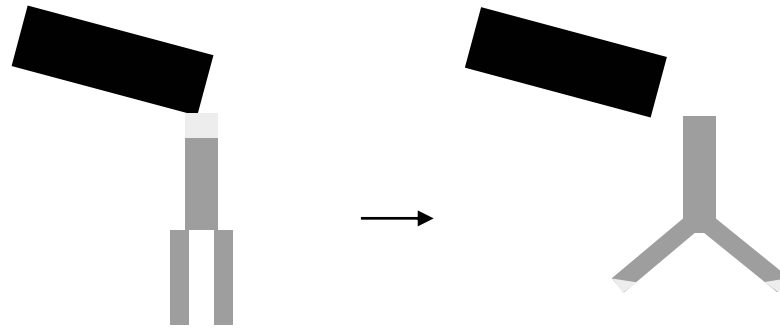


# Static Charges & Electroscopes

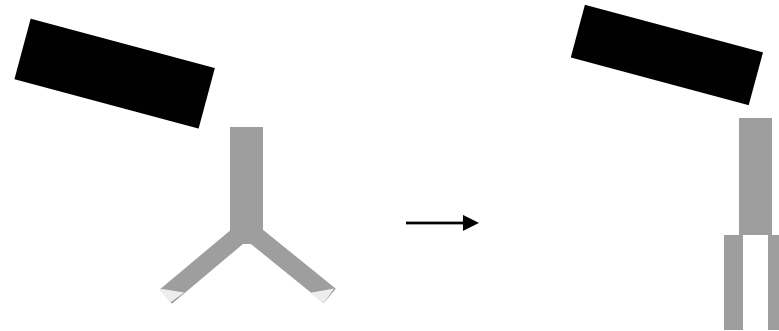
- **The Phenomena**
  - **Charge electroscope with rubber rod which has been rubbed with fur. Gold leaves separate.**
    - » **Bring same rubber rod close to top of electroscope. observe leaves separate further.**
    - » **Bring lucite rod (rubbed with silk) close to top of electroscope. observe leaves approach each other.**
  - **Now repeat experiment, but charge with lucite rod. Gold leaves still separate.**
    - » **Now rubber rod causes leaves to approach each other.**
    - » **Lucite rod causes leaves to separate.**

## More about charging electroscopes (continued)

- **Add some charge (lucite)**

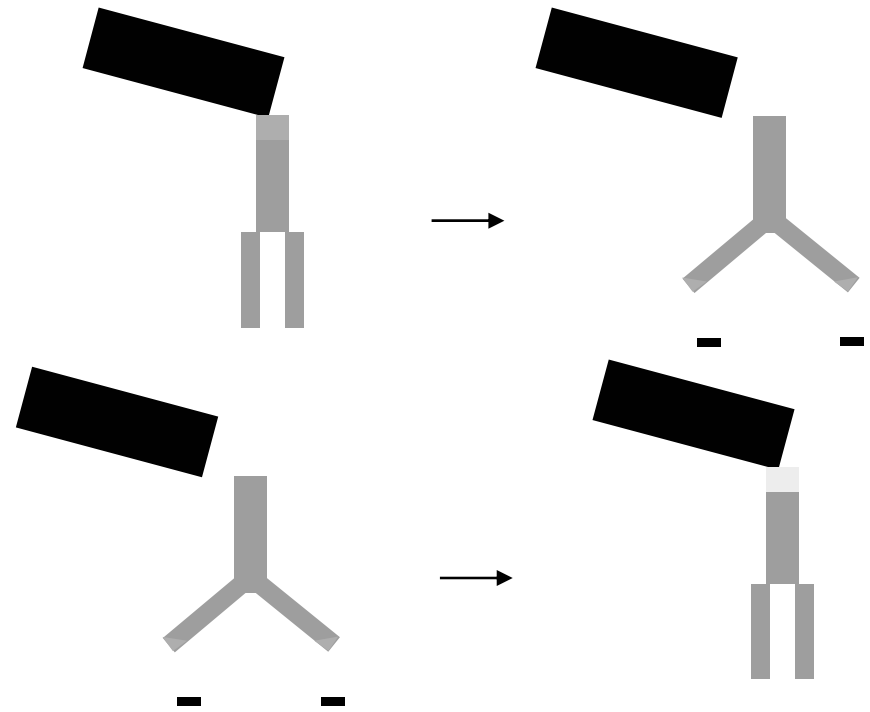
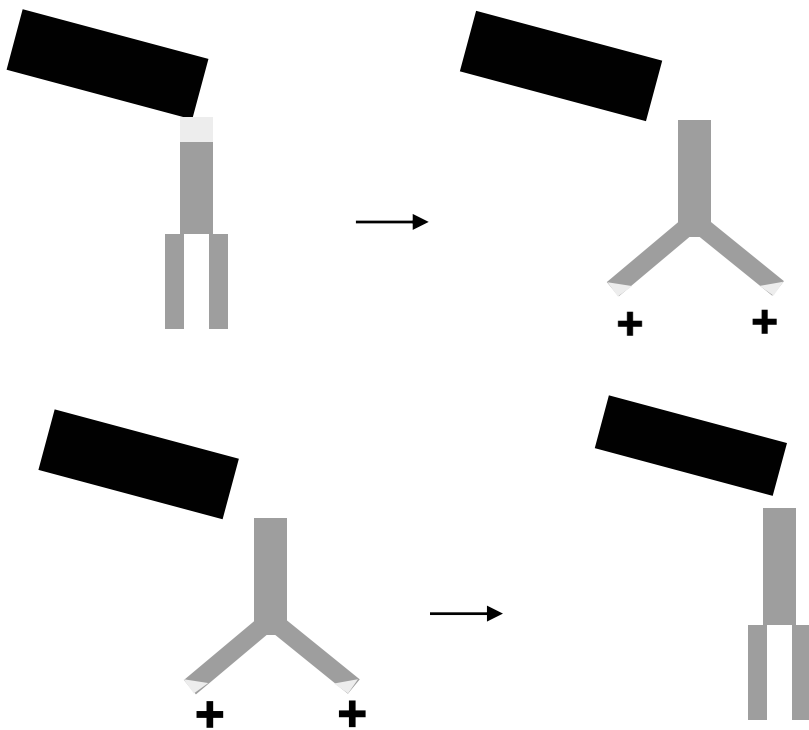


- **Then add charge from rubber to lucite charged leaves**



# Electroscope Conundrum

- Rubber charge
- Lucite charge
  - Opposite of rubber!



- And start with lucite charge...
- Which gets cancelled by rubber charge!
- Hey, we could call these plus and minus!



# Static Charges & Electroscopes

- **The Phenomena**
  - **Charge electroscope with rubber rod which has been rubbed with fur. Gold leaves separate.**
    - » **Bring same rubber rod close to top of electroscope. observe leaves separate further.**
    - » **Bring lucite rod (rubbed with silk) close to top of electroscope. observe leaves approach each other.**
  - **Now repeat experiment, but charge with lucite rod. Gold leaves still separate.**
    - » **Now rubber rod causes leaves to approach each other.**
    - » **Lucite rod causes leaves to separate.**
- **What have we learned?**
  - **Similar charges repel (leaves separate)**
  - **There exist two kinds of charge (lucite negates rubber and vice versa)**

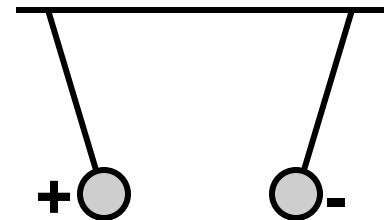
**Text Reference: Chapter 21.1 → 21.5**

# What is this “charge”?

- The charge we see in these experiments comes from atoms: electrons and ions
  - Friction (rubbing) separates them
  - Two types: positive and negative (Ben Franklin!)
    - Positive is charge on rubbed lucite
    - Negative is charge we see on plastic or rubber

- As we saw, like charges repel

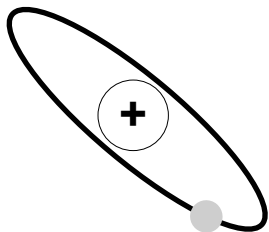
- Opposite charges attract! See pith balls



- Atomic charges are huge! ( $N_A \sim 6 \times 10^{23}$ )

- Electron charge is  $-1.6 \times 10^{-19}$  Coulombs

- Implies that ionized atoms are positive

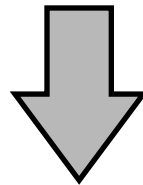


# **Law of Electrical Force**

## **Charles-Augustin Coulomb**

### **1785**

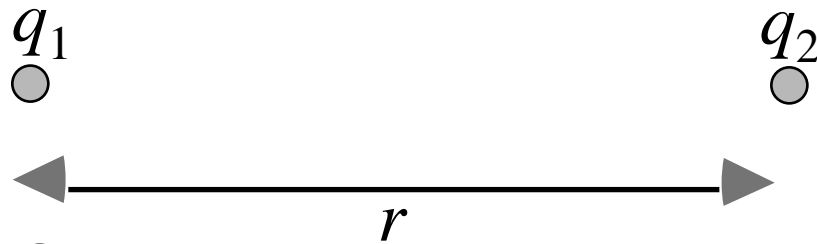
- **"Construction and use of an electric balance - based on the properties of metallic wires of having a force of reaction of torsion proportional to the angle of torsion."**



- **" The repulsive force between two small spheres charged with the same sort of electricity is in the inverse ratio of the squares of the distances between the centers of the spheres"**

# What We Call Coulomb's Law

*The force from  
1 acting on 2*



$$\vec{F}_{12} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r_{12}^2} \hat{r}_{12}$$

**MKS Units:**

- $r$  in meters
- $q$  in Coulombs
- $\vec{F}$  in Newtons

$$\Rightarrow \frac{1}{4\pi\epsilon_0} = 9 \cdot 10^9 \text{ N}\cdot\text{m}^2/\text{C}^2 = k$$

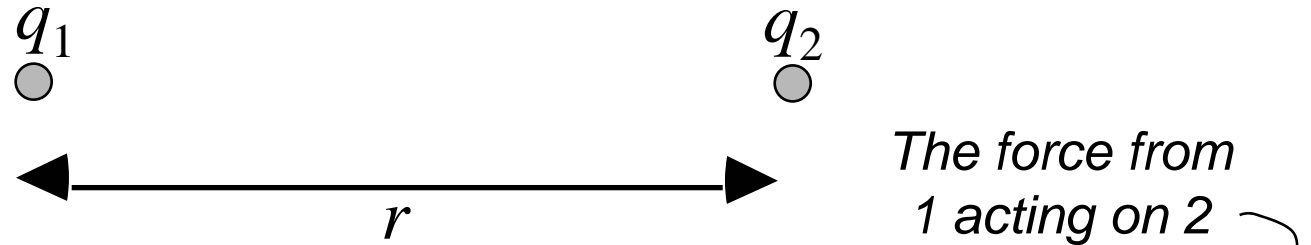
We call this group of constants “ $k$ ” as  
in:  $\vec{F} = k \frac{q_1 q_2}{r^2} \hat{r}$

**This force has same spatial dependence as the gravitational force,  
BUT there is NO mention of mass here!!**

**The strength of the FORCE between two objects is determined by the  
charge of the two objects, and the separation between them.**

**The direction of the FORCE is along the  
line of separation. “Central force”**

# Coulomb Law Qualitative



$$\vec{F}_{12} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r_{12}^2} \hat{r}_{12}$$

- What happens if  $q_1$  increases?

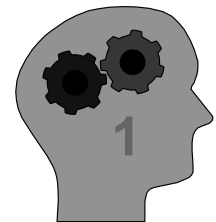
$F$  (magnitude) increases

- What happens if  $q_1$  changes sign ( +  $\longrightarrow$  - )?

The direction of  $\vec{F}$  is reversed

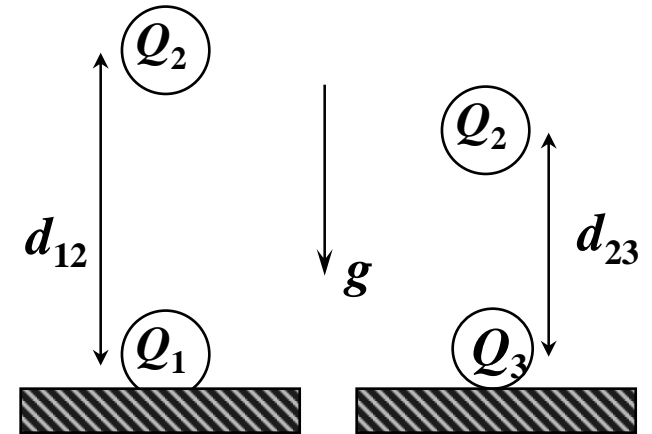
- What happens if  $r$  increases?

$F$  (magnitude) decreases



# Lecture 1, Concept 1

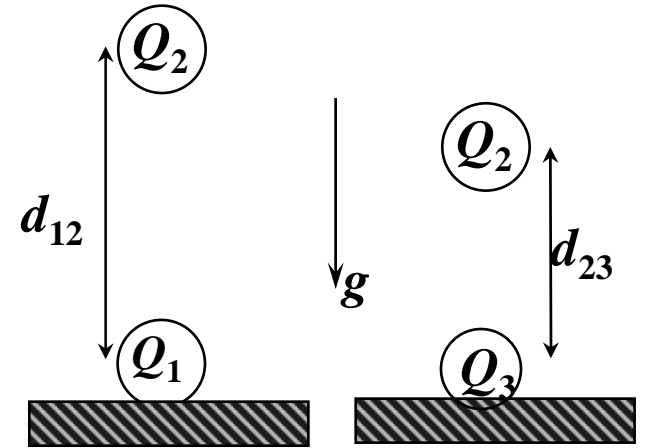
- A charged ball  $Q_1$  is fixed to a horizontal surface as shown. When another massive charged ball  $Q_2$  is brought near, it achieves an equilibrium position at a distance  $d_{12}$  directly above  $Q_1$ .
- When  $Q_1$  is replaced by a different charged ball  $Q_3$ ,  $Q_2$  achieves an equilibrium position at distance  $d_{23}$  ( $< d_{12}$ ) directly above  $Q_3$ .



- 1a:
- A) The charge of  $Q_3$  has the same sign of the charge of  $Q_1$
  - B) The charge of  $Q_3$  has the opposite sign as the charge of  $Q_1$
  - C) Cannot determine the relative signs of the charges of  $Q_3$  &  $Q_1$
- 1b:
- A) The magnitude of charge  $Q_3 <$  the magnitude of charge  $Q_1$
  - B) The magnitude of charge  $Q_3 >$  the magnitude of charge  $Q_1$
  - C) Cannot determine relative magnitudes of charges of  $Q_3$  &  $Q_1$

# Lecture 1, Concept 1

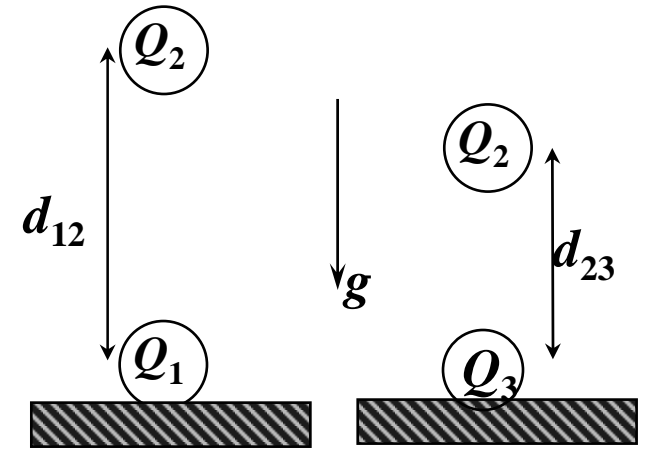
- A charged ball  $Q_1$  is fixed to a horizontal surface as shown. When another massive charged ball  $Q_2$  is brought near, it achieves an equilibrium position at a distance  $d_{12}$  directly above  $Q_1$ .
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- 1a: A) The charge of  $Q_3$  has the same sign of the charge of  $Q_1$
- B) The charge of  $Q_3$  has the opposite sign as the charge of  $Q_1$
- C) Cannot determine the relative signs of the charges of  $Q_3$  &  $Q_1$
- To be in equilibrium, the total force on  $Q_2$  must be zero.
  - The only known (from 121) force acting on  $Q_2$  is its weight.
  - Therefore, in both cases, the electrical force on  $Q_2$  must be directed upward to cancel its weight.
  - Therefore, the sign of  $Q_3$  must be the SAME as the sign of  $Q_1$

# Lecture 1, Concept 1

- A charged ball  $Q_1$  is fixed to a horizontal surface as shown. When another massive charged ball  $Q_2$  is brought near, it achieves an equilibrium position at a distance  $d_{12}$  directly above  $Q_1$ .
- When  $Q_1$  is replaced by a different charged ball  $Q_3$ ,  $Q_2$  achieves an equilibrium position at distance  $d_{23}$  ( $< d_{12}$ ) directly above  $Q_3$ .

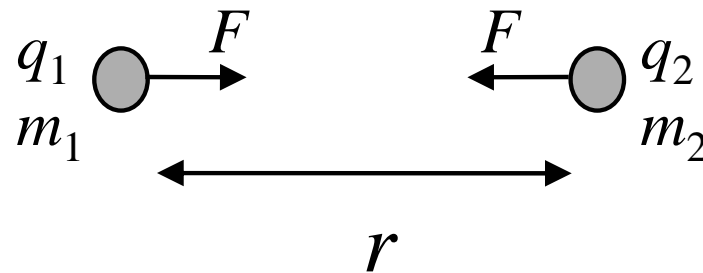


- 1b: A) The magnitude of charge  $Q_3 <$  the magnitude of charge  $Q_1$   
B) The magnitude of charge  $Q_3 >$  the magnitude of charge  $Q_1$   
C) Cannot determine relative magnitudes of charges of  $Q_3$  &  $Q_1$

- The electrical force on  $Q_2$  must be the same in both cases ... it just cancels the weight of  $Q_2$ .
- Since  $d_{23} < d_{12}$ , the charge of  $Q_3$  must be **SMALLER** than the charge of  $Q_1$  so that the total electrical force can be the same!!

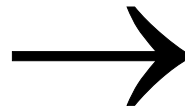


# Gravitational vs. Electrical Force



$$F_{elec} = k \frac{q_1 q_2}{r^2}$$

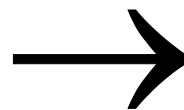
$$F_{grav} = G \frac{m_1 m_2}{r^2}$$



$$\frac{F_{elec}}{F_{grav}} = \frac{q_1 q_2}{m_1 m_2} \frac{k}{G}$$

For a proton:

$$\begin{aligned} * q &= 1.6 \times 10^{-19} \text{ C} \\ m &= 1.67 \times 10^{-27} \text{ kg} \end{aligned}$$

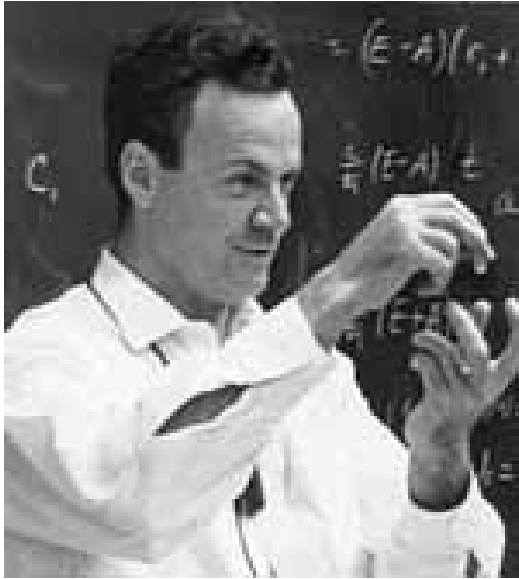


$$\frac{F_{elec}}{F_{grav}} = 1.23 \times 10^{+36}$$

\* Proton charge is smallest charge seen (alone) in nature!

# How Strong is the Electrical Force? Really?

(see Appendix for justification)



**Richard Feynman (1918-1988)**

- **Nobel Prize for QED**
- **Explains Challenger disaster**
- **Educator Extraordinaire**

**Richard Feynman, *The Feynman Lectures*:**

**“If you were standing at arm's length from someone and each of you had *one percent* more electrons than protons, the repelling force would be incredible. How great? Enough to lift the Empire State Building? No! To lift Mount Everest? No! The repulsion would be enough to lift a ‘weight’ equal to that of the entire earth!”**

# Notation For Vectors and Scalars

Vector quantities are written like this :  $\vec{F}, \vec{E}, \hat{x}, \hat{r}$

To completely specify a vector, the magnitude (length) and direction must be known

For example, the following equation shows  $\vec{F}$  specified in terms of  $\hat{r}, q_1, q_2,$  and  $r$

$$\vec{F} = k \frac{q_1 q_2}{r^2} \hat{r}$$

The magnitude of  $\vec{F}$  is  $|\vec{F}| = F = k \frac{q_1 q_2}{r^2}$

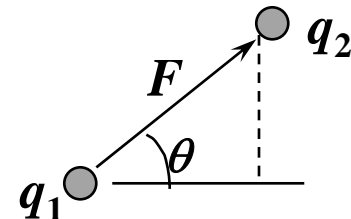
$F$  (the magnitude of  $\vec{F}$ ) is a scalar quantity.

The vector  $\vec{F}$  can be broken down into  $x, y,$  and  $z$  components,

$$\vec{F} = F_x \hat{x} + F_y \hat{y} + F_z \hat{z}$$

where  $F_x, F_y,$  and  $F_z$  (the  $x, y,$  and  $z$  components of  $F$ ) are scalars.

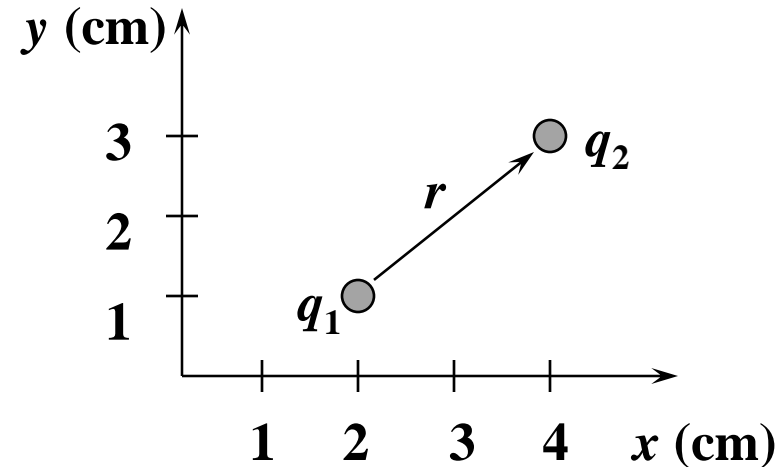
Components of  $\vec{F}$  can be found by *projecting* vector onto an axis



# Vectors and Coulomb's Law: an Example

$q_1$  and  $q_2$  are point charges,  
 $q_1 = +2\mu\text{C}$  and  $q_2 = +3\mu\text{C}$ .  $q_1$  is  
located at  $\vec{r}_1 = (2\text{cm}, 1\text{cm})$  and  
 $q_2$  is located at  $\vec{r}_2 = (4\text{cm}, 3\text{cm})$

Find  $F_{12}$  (the magnitude of the  
force of  $q_1$  on  $q_2$ ). How?



To do this, use Coulomb's Law:  $F_{12} = k \frac{q_1 q_2}{r^2}$

where  $r = |\vec{r}_2 - \vec{r}_1| = \sqrt{(x_2 - x_1)^2 + (y_2 - y_1)^2}$

Now, calculate  $r$ ... How?  $r = \sqrt{(4-2)^2 + (3-1)^2} = \sqrt{8} \text{ cm} = 2.828 \text{ cm}$

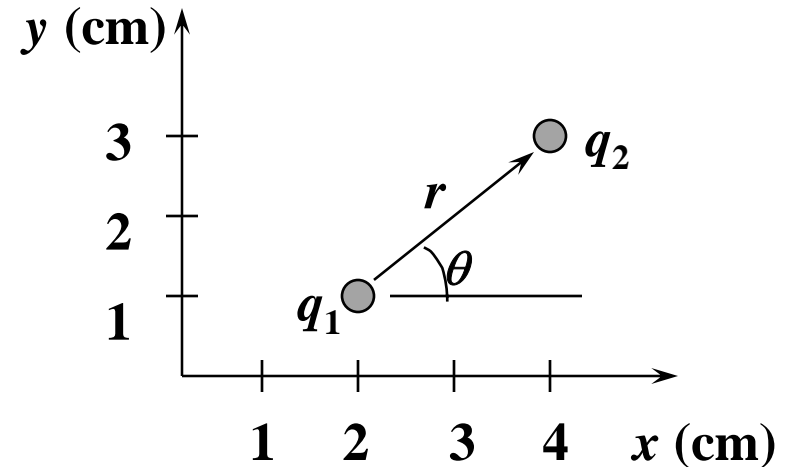
$$F_{12} = k \frac{q_1 q_2}{r^2} = (8.98 \times 10^9 \text{ Nm}^2/\text{C}^2) \frac{(2 \times 10^{-6} \text{ C})(3 \times 10^{-6} \text{ C})}{(0.028)^2 \text{ m}^2} = 68.7 \text{ N}$$

# Vectors: an Example continued

Now, find  $F_x$  and  $F_y$ , the  $x$  and  $y$  components of the force of  $q_1$  on  $q_2$

Direction of force?

Two positive charges repel!



Project Components of Vectors

Got Trigonometry?

$$F_x = F_{12} \frac{x_2 - x_1}{r}$$

← Span in x
← Total length

$$F_y = F_{12} \frac{y_2 - y_1}{r}$$

$$\cos \theta = \frac{x_2 - x_1}{r}$$

$$\sin \theta = \frac{y_2 - y_1}{r}$$



# Vectors: an Example continued

Now, find  $F_x$  and  $F_y$ , the  $x$  and  $y$  components of the force of  $q_1$  on  $q_2$

## Components of Vectors

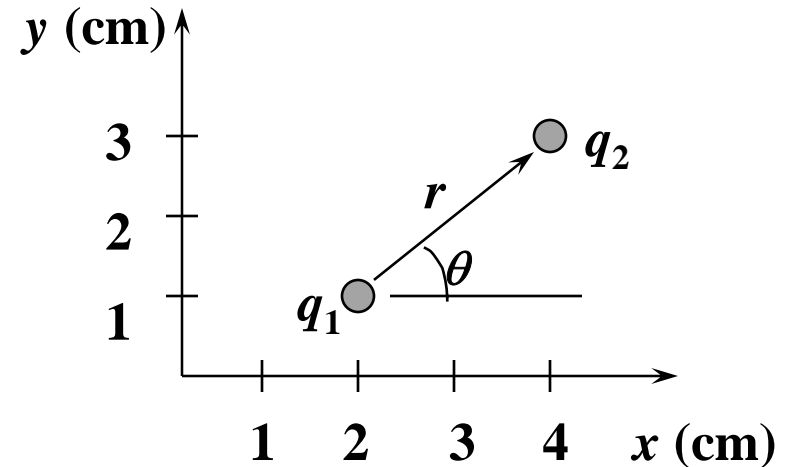
$$F_x = F_{12} \frac{x_2 - x_1}{r} \quad \begin{array}{l} \leftarrow \text{Span in } x \\ \leftarrow \text{Total length} \end{array}$$

$$F_y = F_{12} \frac{y_2 - y_1}{r}$$

$$F_{12} = k \frac{q_1 q_2}{r^2} = 68.7 \text{ N}$$

$$\frac{x_2 - x_1}{r} = \frac{4 - 2}{\sqrt{8}} = .707 \quad F_x = 47.74 \text{ N}$$

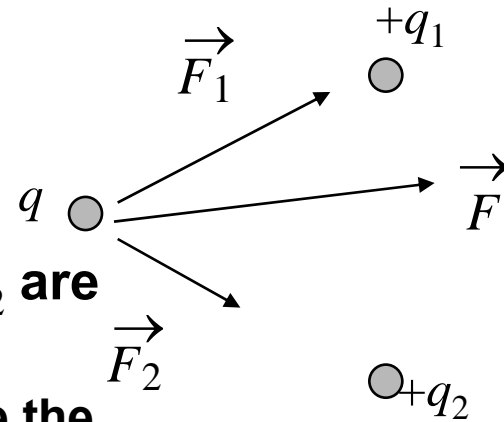
$$\frac{y_2 - y_1}{r} = \frac{3 - 1}{\sqrt{8}} = .707 \quad F_y = 47.74 \text{ N}$$



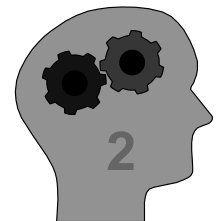
What are those factors numerically?

# What happens when you consider more than two charges?

- If  $q_1$  were the only other charge, we would know the force on  $q$  due to  $q_1$ .
- If  $q_2$  were the only other charge, we would know the force on  $q$  due to  $q_2$ .
- What is the force on  $q$  when both  $q_1$  and  $q_2$  are present??
  - The answer: just as in mechanics, we have the **Law of Superposition**:
    - The **TOTAL FORCE** on the object is just the **VECTOR SUM** of the individual forces.

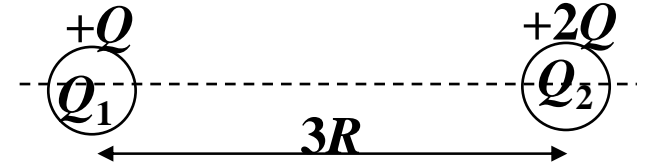


$$\vec{F} = \vec{F}_1 + \vec{F}_2$$

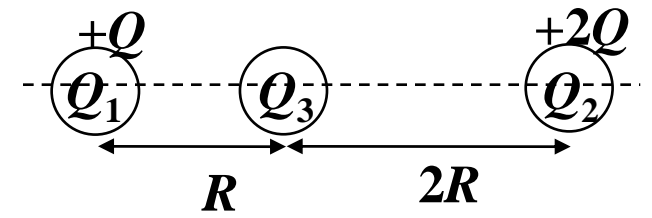


# Lecture 1, Concept 2

- Two balls, one with charge  $Q_1 = +Q$  and the other with charge  $Q_2 = +2Q$ , are held fixed at a separation  $d = 3R$  as shown.



- Another ball with (non-zero) charge  $Q_3$  is introduced in between  $Q_1$  and  $Q_2$  at a distance  $= R$  from  $Q_1$ .
- Which of the following statements is true?

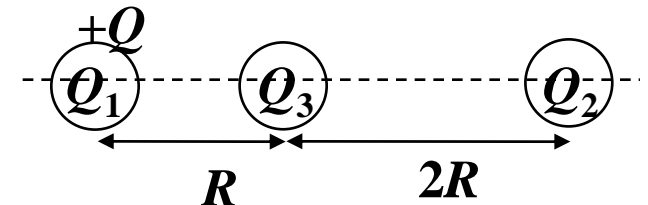
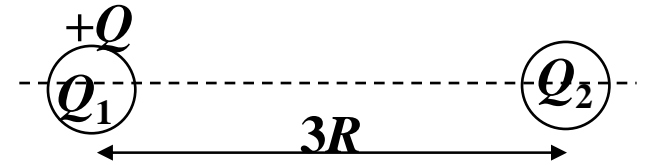


- (a) The net force on  $Q_3$  can be zero if  $Q_3$  is positive.
- (b) The net force on  $Q_3$  can be zero if  $Q_3$  is negative.
- (c) The net force on  $Q_3$  can never be zero, no matter what the (non-zero!) charge  $Q_3$  is.



# Lecture 1, Concept 2

- Two balls, one with charge  $Q_1 = +Q$  and the other with charge  $Q_2 = +2Q$ , are held fixed at a separation  $d = 3R$  as shown.
- Another ball with (non-zero) charge  $Q_3$  is introduced in between  $Q_1$  and  $Q_2$  at a distance  $= R$  from  $Q_1$ .
  - Which of the following statements is true?



- (a) The force on  $Q_3$  can be zero if  $Q_3$  is positive.
- (b) The force on  $Q_3$  can be zero if  $Q_3$  is negative.
- (c) The force on  $Q_3$  can never be zero, no matter what the (non-zero) charge  $Q_3$  is.**

The magnitude of the force on  $Q_3$  due to  $Q_2$  is proportional to  $(2Q Q_3 / (2R)^2)$

The magnitude of the force on  $Q_3$  due to  $Q_1$  is proportional to  $(Q Q_3 / R^2)$

These forces can never cancel, because the force  $Q_2$  exerts on  $Q_3$  will always be 1/2 of the force  $Q_1$  exerts on  $Q_3$ !!

# Today's Summary

- **Charges come in two varieties**
  - negative and positive
  - like charges repel but opposites attract

- **Coulomb Force**

- linear in both charges
- inversely proportional to square of separation
- central force

$$\vec{F}_{12} = k \frac{q_1 q_2}{r_{12}^2} \hat{r}_{12}$$

- **Law of Superposition**

$$\vec{F} = \vec{F}_1 + \vec{F}_2$$

*Reading assignment: (21.1-5 review) 21.6, 21.8-9, 21.11  
study examples 21.2, 21.3, 21.6, 21.7, 21.16*

# Appendix B: Should we believe Feynman?

- **How many electrons in a person?**

- What do we assume is the chemical composition of a person?

Simplify: assume water (molecular weight = 18)

- What then is the number of electrons/gram in a person?

$$\frac{6 \times 10^{23} \text{ molecules/mole}}{18 \text{ g/mole}} \times 10 \text{ e}^-/\text{molecule} = 3.3 \times 10^{23} \text{ e}^-/\text{g}$$

- So, how many electrons in a person?

$$\text{Assume weight} = 80 \text{ kg} \Rightarrow 3.3 \times 10^{23} \text{ e}^-/\text{g} \times 80 \text{ kg} = 2.6 \times 10^{28} \text{ e}^-$$

- How much charge is 1% of electrons in a person?

$$1\% \times 2.6 \times 10^{28} \text{ e}^- \times 1.6 \times 10^{-19} \text{ C/e}^- = 4.2 \times 10^7 \text{ C}$$

# Should we believe?

- What is the force between 2 people an arm's length apart if they each had an excess of 1% electrons?

$$F = (9 \times 10^9 \text{ N}\cdot\text{m}^2/\text{C}^2) \times \left( \frac{4.2 \times 10^7 \text{ C}}{0.75 \text{ m}} \right)^2$$

$$F = 2.8 \times 10^{25} \text{ N}$$

- What is the “weight” of the earth?

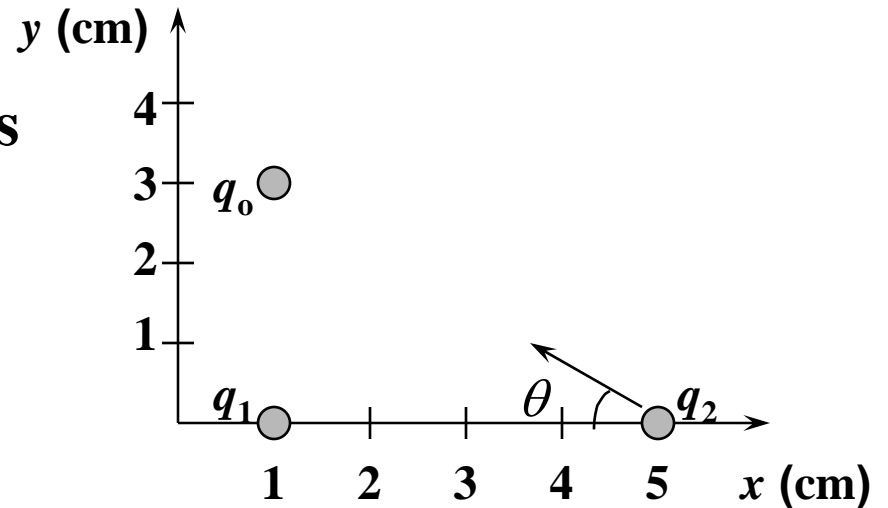
$$W_{\text{earth}} = 6 \times 10^{24} \text{ kg} \times 9.8 \text{ m/s}^2$$

$$W_{\text{earth}} = 5.9 \times 10^{25} \text{ N}$$

- Yes, that's INCREDIBLE!!

# Appendix C: A Longer Example

$q_0$ ,  $q_1$ , and  $q_2$  are all point charges where  $q_0 = -1\mu\text{C}$ ,  $q_1 = 3\mu\text{C}$ , and  $q_2 = 4\mu\text{C}$ . Their locations are shown in the diagram. What is the force acting on  $q_0$ ? ( $\vec{F}_0$ )



**Superposition:**

Find  $\vec{F}_{10}$  and  $\vec{F}_{20}$

$$F_{10} = \left| k \frac{q_0 q_1}{r_{10}^2} \right| \quad F_{20} = \left| k \frac{q_0 q_2}{r_{20}^2} \right|$$

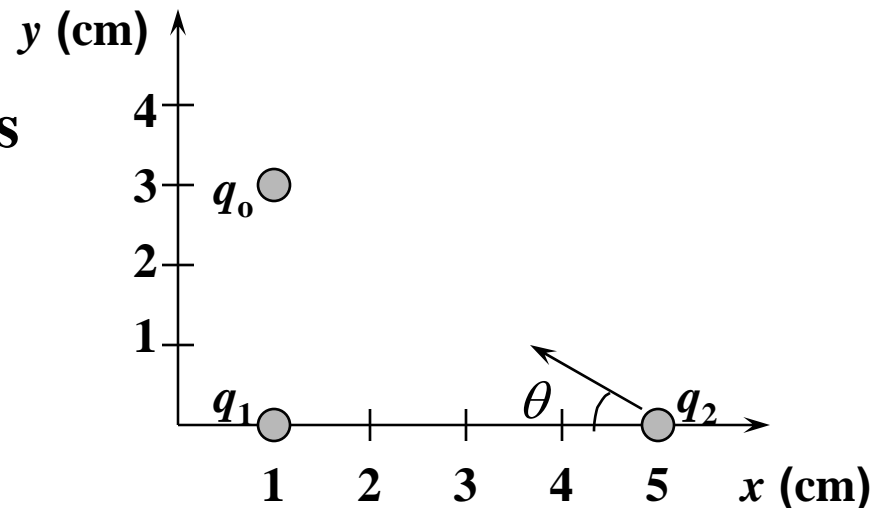
**Direction?**

$$\vec{F}_{10} = -F_{10} \hat{y} \quad \vec{F}_{20} = -F_{20} \hat{r}_{20}$$

**Decompose  $\vec{F}_{20}$  into its  $x$  and  $y$  components**

# A Longer Example

$q_0$ ,  $q_1$ , and  $q_2$  are all point charges where  $q_0 = -1\mu\text{C}$ ,  $q_1 = 3\mu\text{C}$ , and  $q_2 = 4\mu\text{C}$ . Their locations are shown in the diagram. What is the force acting on  $q_0$ ? ( $\vec{F}_0$ )



Decompose  $\vec{F}_{20}$  into its  $x$  and  $y$  components

$$\vec{F}_{20} = F_{20} \frac{x_2 - x_0}{r_{20}} \hat{x} + F_{20} \frac{y_2 - y_0}{r_{20}} \hat{y}$$

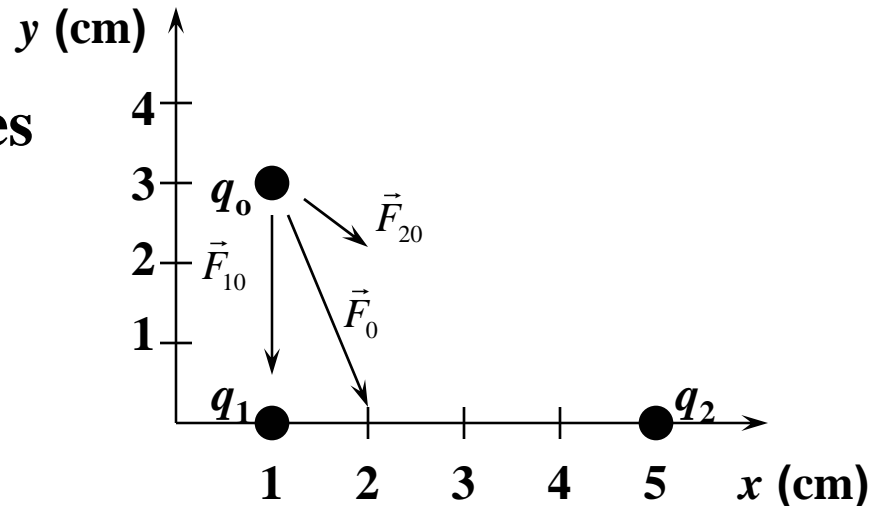
← Span in x
← Total length

Superimposing forces...

$$\vec{F}_0 = \vec{F}_{10} + \vec{F}_{20} = F_{20} \frac{x_2 - x_0}{r_{20}} \hat{x} + \left( F_{20} \frac{y_2 - y_0}{r_{20}} - F_{10} \right) \hat{y}$$

# A Longer Example continued

$q_0$ ,  $q_1$ , and  $q_2$  are all point charges where  $q_0 = -1\mu\text{C}$ ,  $q_1 = 3\mu\text{C}$ , and  $q_2 = 4\mu\text{C}$ . Their locations are shown in the diagram. What is the force acting on  $q_0$ ? ( $\vec{F}_0$ )



Let's put in the numbers . . .

$$r_{10} = 3\text{cm}$$

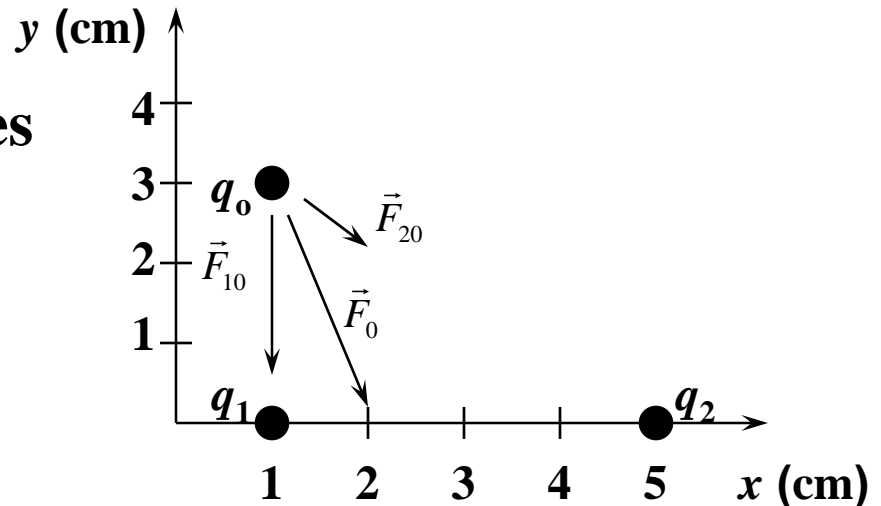
$$r_{20} = \sqrt{(x_2 - x_0)^2 + (y_2 - y_0)^2} = \sqrt{4^2 + (-3)^2} \text{cm} = 5\text{cm}$$

$$F_{10} = k \frac{|q_0 q_1|}{r^2} = (8.98 \times 10^9 \text{ Nm}^2/\text{C}^2) \frac{(1 \times 10^{-6} \text{ C})(3 \times 10^{-6} \text{ C})}{(0.03)^2 \text{ m}^2} = 29.9 \text{ N}$$

$$F_{20} = k \frac{|q_0 q_2|}{r_{20}^2} = (8.98 \times 10^9 \text{ Nm}^2/\text{C}^2) \frac{(1 \times 10^{-6} \text{ C})(4 \times 10^{-6} \text{ C})}{(0.05)^2 \text{ m}^2} = 14.4 \text{ N}$$

# A Longer Example continued

$q_0$ ,  $q_1$ , and  $q_2$  are all point charges where  $q_0 = -1\mu\text{C}$ ,  $q_1 = 3\mu\text{C}$ , and  $q_2 = 4\mu\text{C}$ . Their locations are shown in the diagram. What is the force acting on  $q_0$ ? ( $\vec{F}_0$ )



Let's put in the numbers . . .

$$r_{10} = 3\text{cm} \qquad F_{10} = 29.9\text{N}$$

$$r_{20} = 5\text{cm} \qquad F_{20} = 14.4\text{N}$$

$$\begin{aligned} \vec{F}_0 &= F_{20} \frac{x_2 - x_0}{r_{20}} \hat{x} + \left( F_{20} \frac{y_2 - y_0}{r_{20}} - F_{10} \right) \hat{y} \\ &= F_{20} \frac{(5 - 1)\text{cm}}{5\text{cm}} \hat{x} + \left( F_{20} \frac{(0 - 3)\text{cm}}{5\text{cm}} - F_{10} \right) \hat{y} \\ &= (11.5 \hat{x} - 38.5 \hat{y}) \text{N} \end{aligned}$$