

# Resolving Power

of a grating  
→ spectrometry



input light

Chromatic Resolving Power

$$R = \frac{\lambda}{\Delta\lambda} = Nm$$

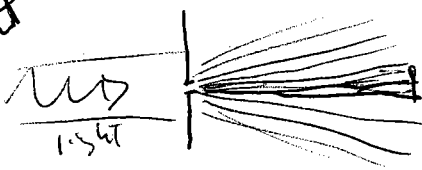
$$\lambda \sim \lambda_1 \sim \lambda_2 \sim \frac{\lambda_1 + \lambda_2}{2}$$

$$\Delta\lambda = \lambda_2 - \lambda_1$$

Resolution limit comes when angular separation of 2 wavelengths = distance between Maximum and 1<sup>st</sup> minimum for the wavelengths

## Resolution of point objects - Rayleigh's criterion

of an optical  
instrument



Diffraction

image

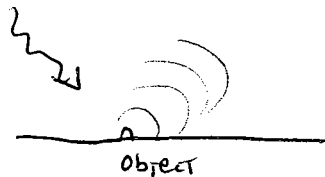
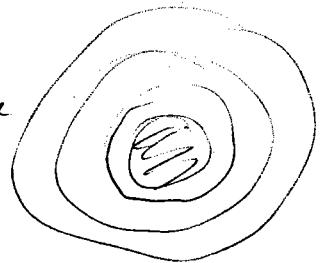


image of small object ~~can~~ is "diffraction limited" when size  $\sim \lambda$

If The Angular separation of two point sources is ~~just res~~ such that the Maximum for one diffraction pattern falls on the 1<sup>st</sup> minimum of the other - The two point sources are at the threshold for resolvability

$$\theta_R = 1.22 \frac{\lambda}{d} \quad \text{Rayleigh's criterion}$$

# ⇒ Do Young's Double Slit Demos

Easter Egg candy to whoever can answer

⇒ Why do camera lenses look dark?

⇒ Do soap film Demos

↑ How does this work?

$$c = \lambda v$$

$$v = \lambda n v$$

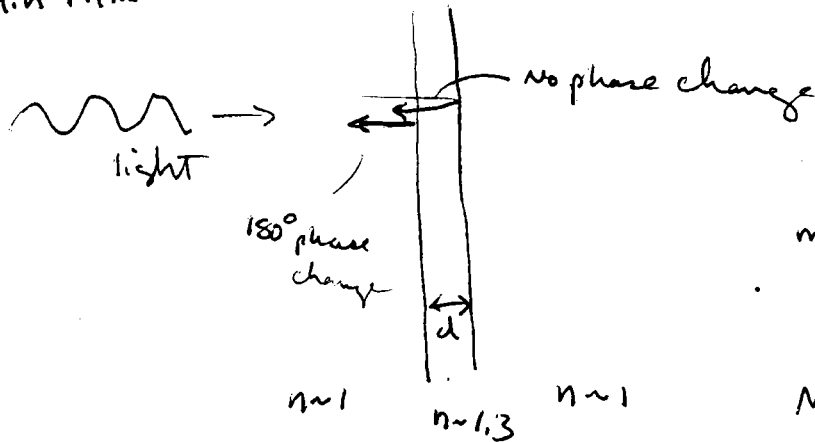
$$v n = c$$

$$\frac{c}{n} = \lambda n v$$

$$\lambda = \frac{c}{n^2 v} = \lambda / n$$

$\lambda$  in medium

## Thin Film



minimum in reflection if

$$2d = m \lambda_n$$

( $m = 1, 2, 3, \dots$ )

Maximum if

$$2d = (m + \frac{1}{2}) \lambda_n$$

( $m = 0, 1, 2, \dots$ )

⇒ light ~~reflects~~ going from low  $n$  to high  $n$

⇒ 180° phase change upon reflection at surface

⇒ light going from high  $n$  to low  $n$

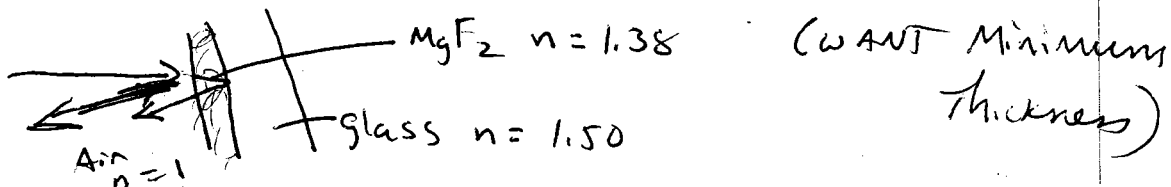
⇒ No phase change upon reflection at surface

No phase change in transmitted wave in either case.

## example

Lenses often coated w/  $MgF_2$  magnesium Fluoride  
to reduce the reflection in the visible spectrum  
(central  $\lambda = 550 \text{ nm}$ )

How Thick should the coating be to minimize reflection at 550 nm



at both interfaces there is a  $180^\circ$  phase change upon reflection.

$\therefore$  interference determined solely by path difference

Minimum if  $2d = (m + \frac{1}{2}) \lambda_n$   
 $(m = 0, 1, 2, \dots)$

$$\lambda_n = \lambda / n$$

Thickness of thinnest possible film

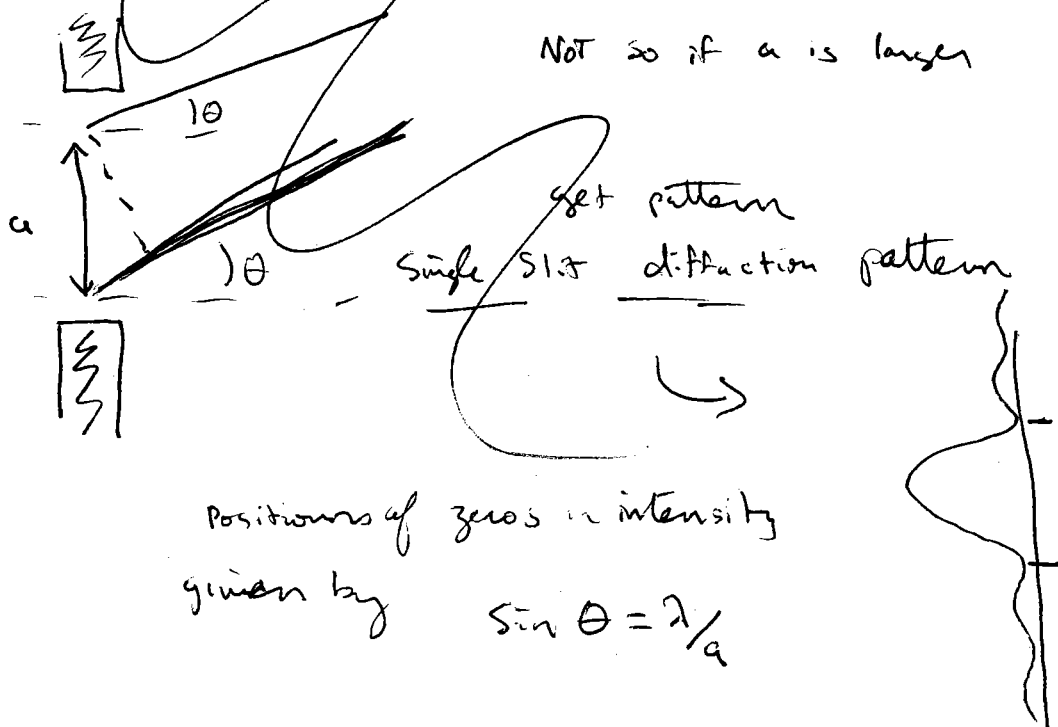
$$2d = (m + \frac{1}{2}) \lambda / n$$

$$d = \frac{\lambda}{4n} = \frac{550 \text{ nm}}{(4)(1.38)} = 100 \text{ nm}$$

## Diffraction Pattern

double slit ~~interference~~ interference discussion

Assumed slits were thin, i.e. all light coming from slit had same path length to screen



at both interfaces there is a  $180^\circ$  phase change upon reflection.

interference determined solely by path difference

Minimum if  $2d = (m + \frac{1}{2}) \lambda_n$   
 ( $m = 0, 1, 2, \dots$ )

$\lambda_n = \lambda / n$

Thickness of thinnest possible film

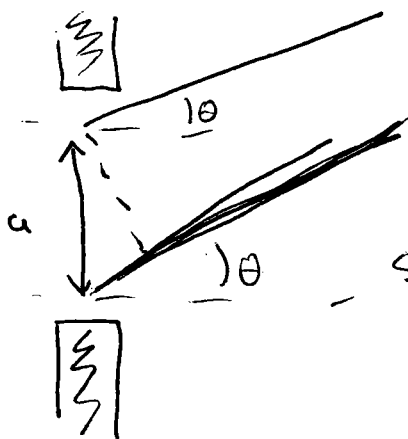
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## Diffraction Pattern

double slit ~~interference~~ interference discussion

Assumed slits were thin, i.e. all light coming from slit had same path length to screen



Not so if  $a$  is larger

get pattern

Single slit diffraction pattern



positions of zeros in intensity given by  $\sin \theta = \lambda / a$



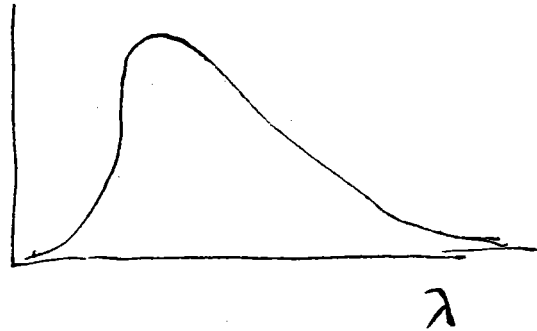
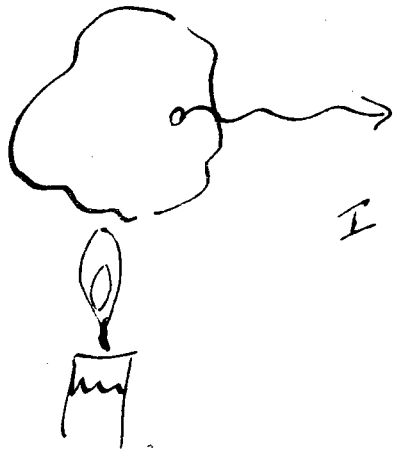
I will ~~not~~ be working from Tipler w/ quite the  
 closeness I have earlier - will be using a different  
 Text next year

3 Things really led to rise of Quantum Mechanics:

(F)

## Black body Radiation

Spectrum of Radiation emitted by an object  
 due solely to its temperature



Classical Treatments failed  
 end of 1800's

In 1900 Max Planck

Worked  
 exactly

$$R(\lambda) \approx I = \frac{c}{4} \left( \frac{8\pi}{\lambda^4} \right) \left[ \frac{hc}{\lambda} \frac{1}{e^{hc/\lambda kT} - 1} \right]$$

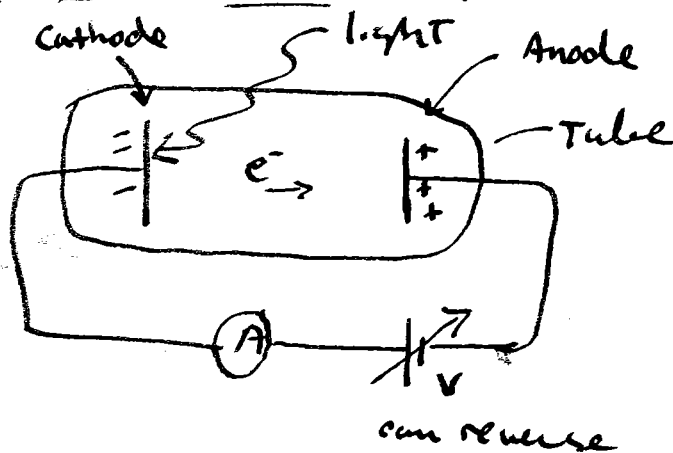
To derive ... Must Assume light comes in little  
 bundles called quanta  $E = h\nu$

Energy of quanta  $\propto \nu$

Proportionality constant  $h$  (Planck's constant)

This was viewed as a curiosity at the time.

## II Photoelectric effect ( $\nu$ )

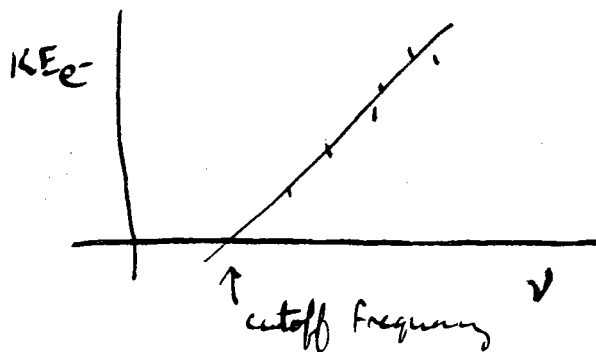


- $e^-$  ejected from metal as light shines
- can vary  $I, \nu$  of light
- can vary  $\nu$  to meas KE of  $e^-$
- Can Measure  $i$  (current) (#  $e^-$  ejected)
- KE of  $e^-$

Classical physics predicted

- Some  $e^-$  ejected w/ any  $\nu$
- double intensity  $\rightarrow$  ~~double~~  $\frac{1}{2}$  time for ejection

Found



1905 Einstein

light  $E = h\nu$

$$KE_{e^-} = h\nu - W$$

work function

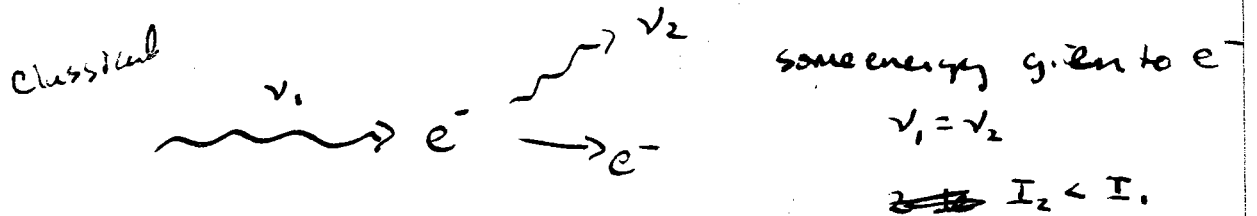
Assumes / uses Planck's quanta

Note  $h$  ~~found~~ photoelectric effect  
could be different from  
 $h$  in black body radiation

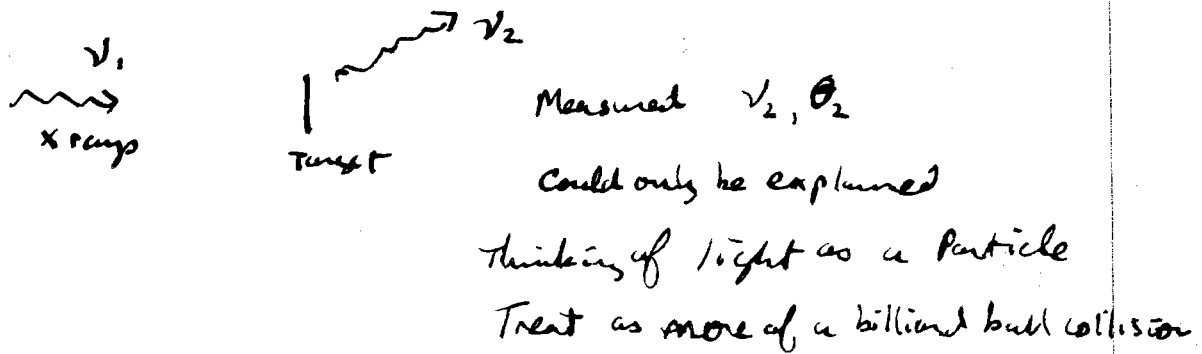
Measure  $h$  experimentally  $\rightarrow$  It is the same!!  
 $h = 6.62 \times 10^{-34}$  J.s

These 2 things started us on the road to quantum theory

III one more thing  
 Compton scattering



1923 Arthur Compton



$\Rightarrow$  There you have it

interference diffraction	}	blackbody rad photoelectric effect Compton Scattering	which is it?
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Wave Particle duality of light

10 SHEETS FILLER 5 SQUARE  
 20 SHEETS FILLER 5 SQUARE  
 30 SHEETS FILLER 5 SQUARE  
 40 SHEETS FILLER 5 SQUARE  
 50 SHEETS FILLER 5 SQUARE  
 60 SHEETS FILLER 5 SQUARE  
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 160 SHEETS FILLER 5 SQUARE  
 170 SHEETS FILLER 5 SQUARE  
 180 SHEETS FILLER 5 SQUARE  
 190 SHEETS FILLER 5 SQUARE  
 200 SHEETS FILLER 5 SQUARE  
 200 RECYCLED WHITE 5 SQUARE  
 MADE IN U.S.A.

National Brand

Things got much STRANGER:

if light waves are particles

Suppose particles are waves

Louis de Broglie ~ 1925 proposed

$$\lambda = \frac{h}{p} \quad \text{de Broglie Wavelength}$$

$h$  is very small

$$\rightarrow \lambda \text{ of } 1000 \text{ Kg Auto travelling at } 100 \text{ m/s} \Rightarrow \lambda = 6.6 \times 10^{-39} \text{ m}$$

$$\rightarrow \lambda \text{ of } e^- \text{ w/ KE of } 1 \text{ electron volt} \rightarrow \lambda = 1.2 \text{ nm}$$

↑  
Small but NOT crazy!

1926 Davisson Germer

observed diffraction of  $e^-$  beam off Nickel!

x ray diffraction

$e^-$  diffraction

neutron diffraction

> crystallography !!

diffraction pattern yields details about  
crystal lattice waves are reflecting from!

So matter also has wave-particle duality!

Particle is a wave ... Should be able to write down a  
Classical wave eqn.

No: We've worked w/ "free" waves

Now we need them to operate in "potentials"

Particles have mass

Must come in quanta  $E = h\nu$ ,  $p = h/\lambda$

Have NOT done that w/ light either