

# Physics 100 - February 12, 2007

■ 12/9 Lecture - notes online

hope to have streaming video available soon  
Will need Realmedia player  
probably need to be in UR domain

■ Exam 1 - Feb 21

Thru  $\sim 1/2$  way 12/9 lect. (marked in notes)  
P.S. 3  
Recitation } + associated reading

■ Project group listings

Maxwell tells us that light is a wave phenomenon.

$$\int \vec{E} \cdot d\vec{a} = \frac{Q_{enc}}{\epsilon_0}$$

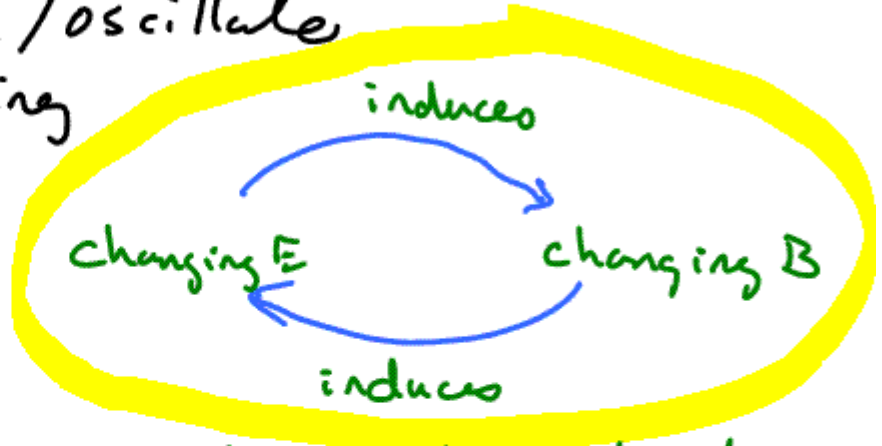
$$\int \vec{B} \cdot d\vec{a} = 0$$

$$\int \vec{E} \cdot d\vec{l} = -\frac{d}{dt} \int \vec{B} \cdot d\vec{a}$$

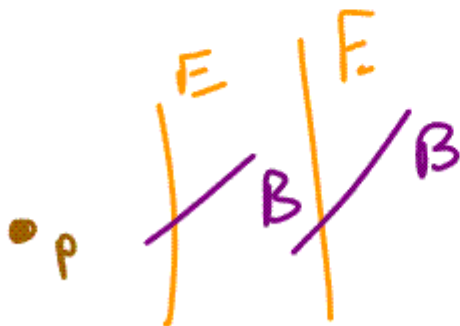
$$\int \vec{B} \cdot d\vec{l} = \mu_0 I_{enc} + \mu_0 \epsilon_0 \frac{d}{dt} \int \vec{E} \cdot d\vec{a}$$

E, B fields wave/oscillate  
Self propagating

E ≡ symbol for electric field  
B ≡ symbol for magnetic field



Fist full of Electric charge → creates changing E which induces changing B which induces changing E ...

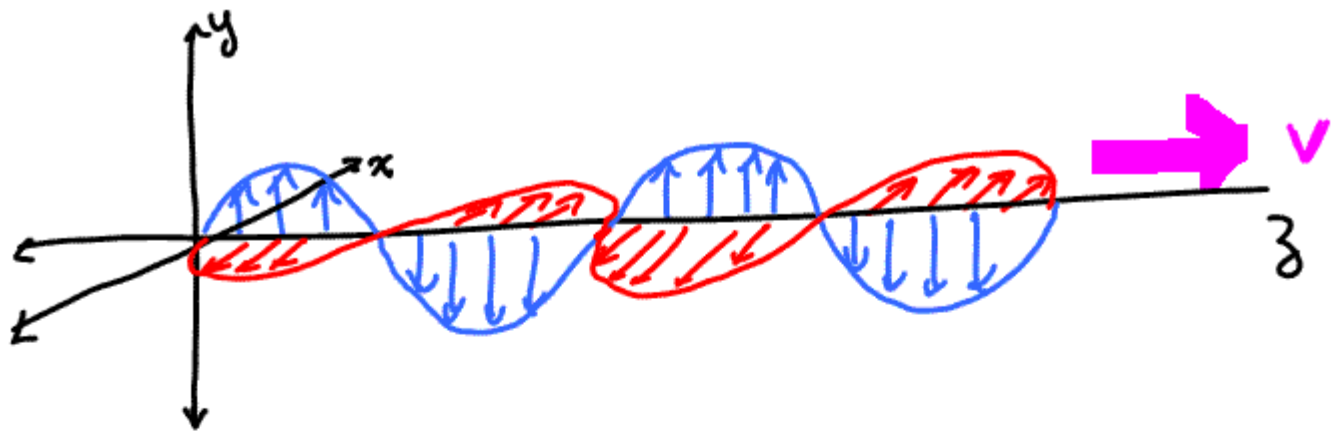


Propagates outward  
at speed of  
light

→ it is light



observer  
very  
far  
away



Wave propagates in  $z$  direction

$\uparrow$  represents direction and magnitude of  $E$  field in plane transverse to motion

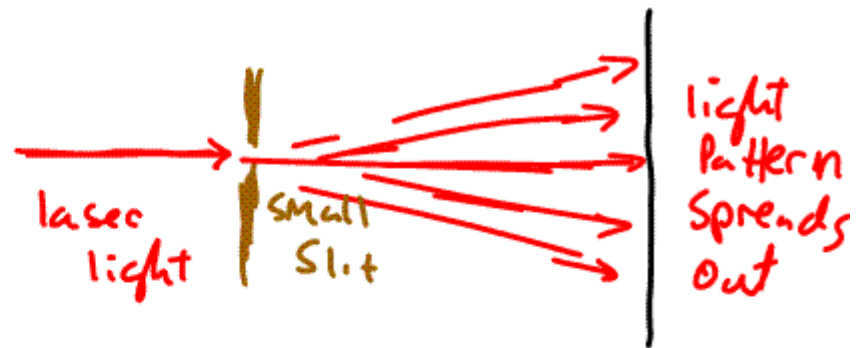
$\rightarrow$  represents direction and magnitude of  $B$  field in plane transverse to direction of motion

Waves

Sound, pulse on string, water, light

All share certain characteristics

# Diffraction - waves spread out passing thru small hole



Period  
 $T \equiv$  Time for  $1\lambda$   
 to pass by fixed pt  
 Time for 1 cork cycle

Frequency  
 $f$  or  $\nu \equiv \frac{1}{T}$

Meas. of  
 energy  
 and  
 intensity

$$v = \lambda \nu$$

$m/s \quad m/s$



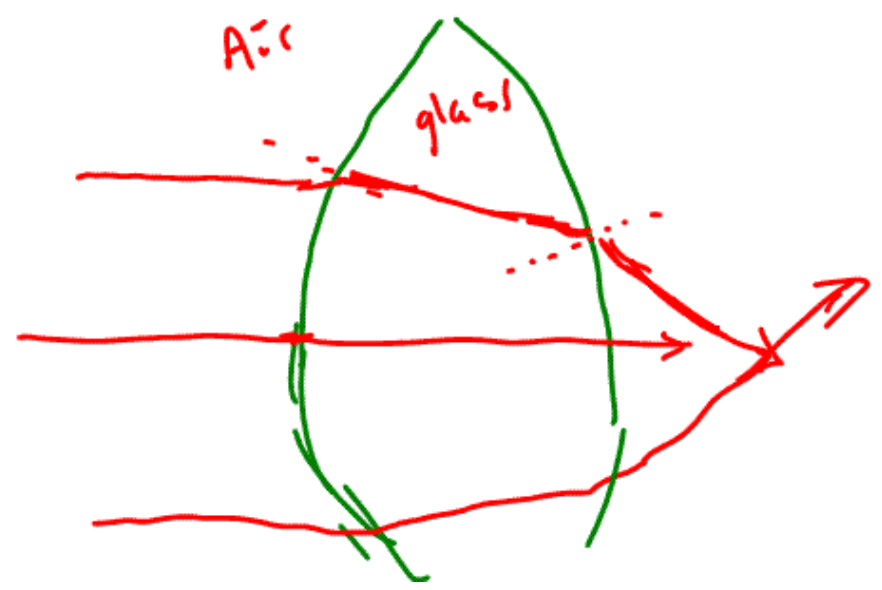
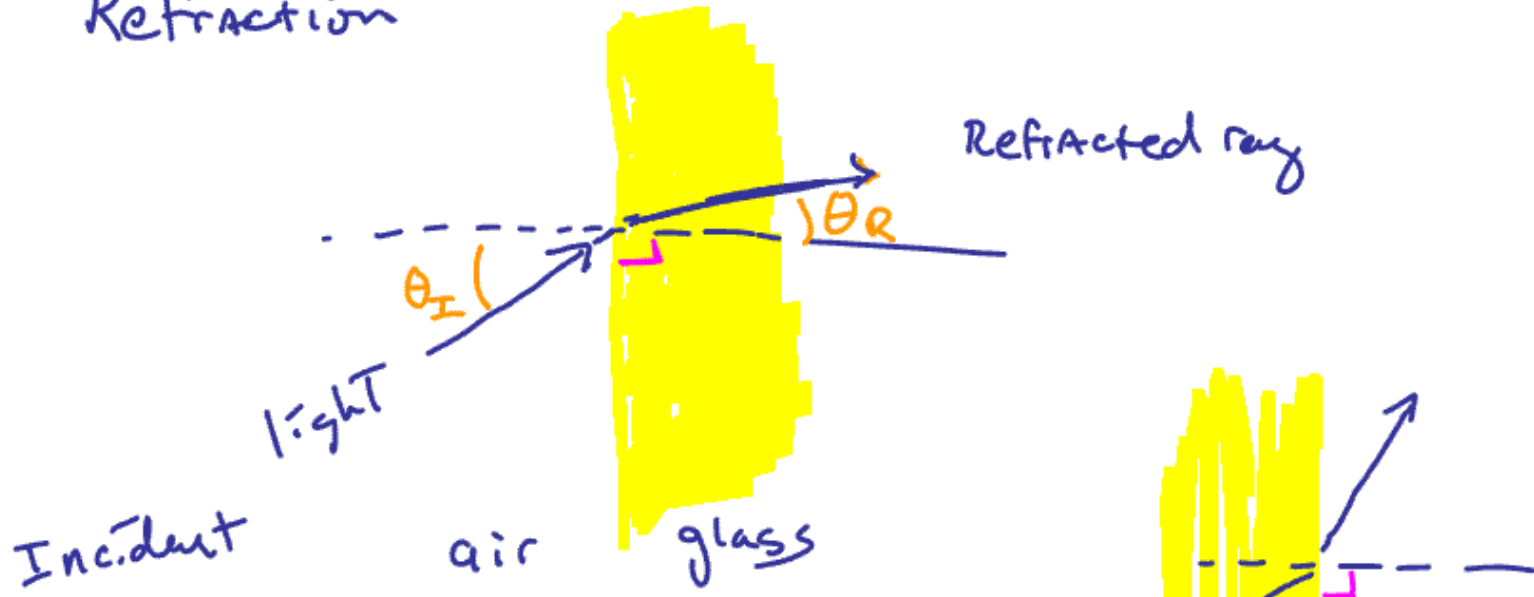
CONSTRUCTIVE  
Interference



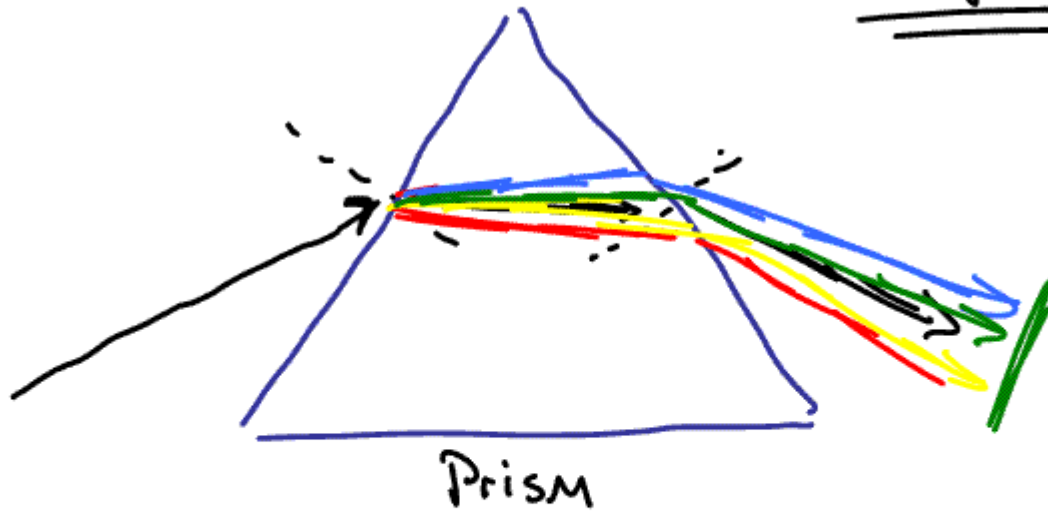
Destructive  
Interference



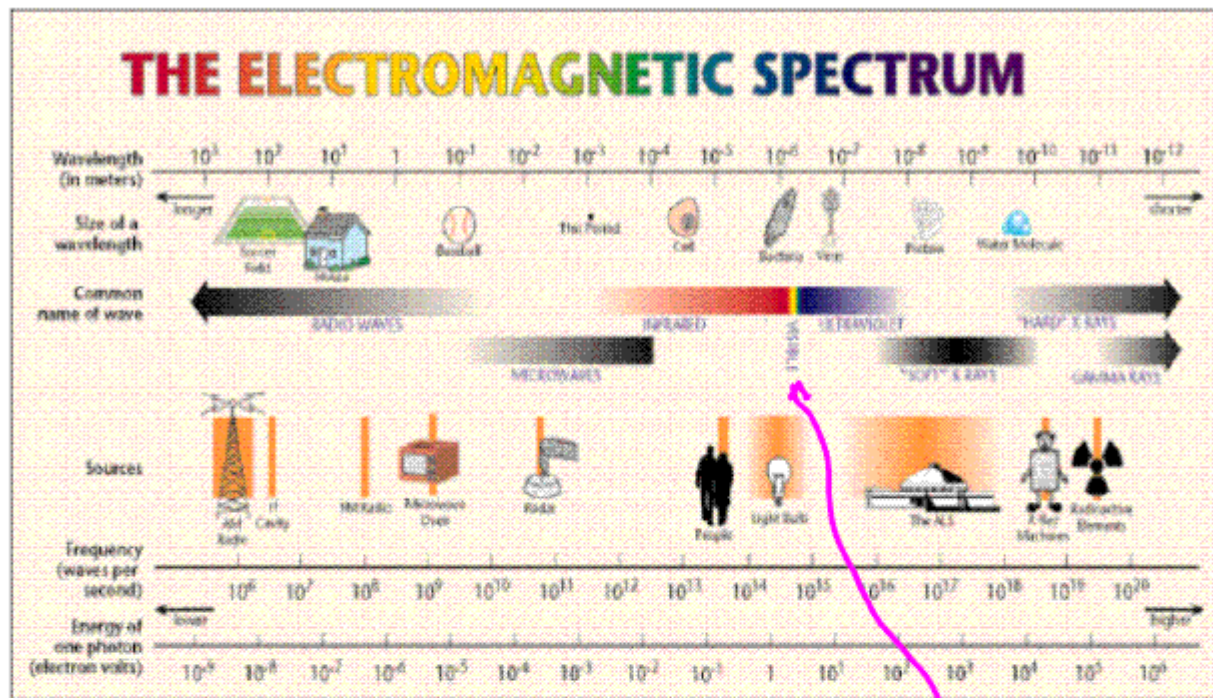
# Refraction



dispersion



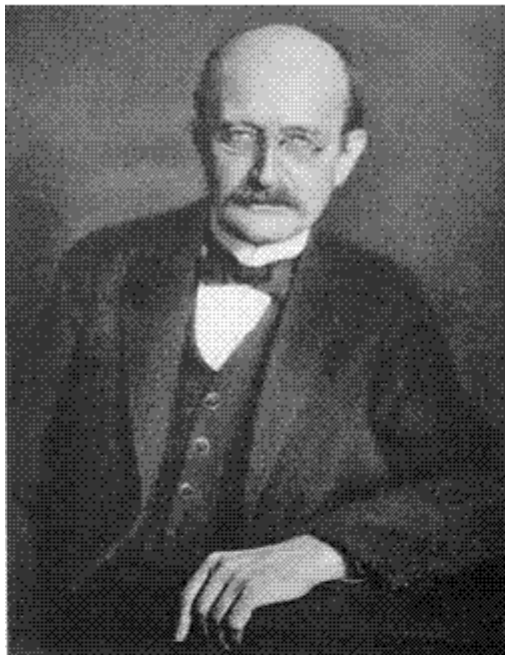
degree by which light is bent  
depends on frequency



- no real limits on wavelength / frequency
  - visible light is but a tiny fraction of "colors" in universe
- visible colors  
frequencies

interaction of light w/ matter is a very fundamental way to study the structure of matter.  
Best to study light emitted by object (not reflected)





# Max Planck

(1858-1947)

German national

Awarded 1918 Nobel Prize in physics  
for analysis of blackbody radiation  
which contributed to rise of  
quantum mechanics

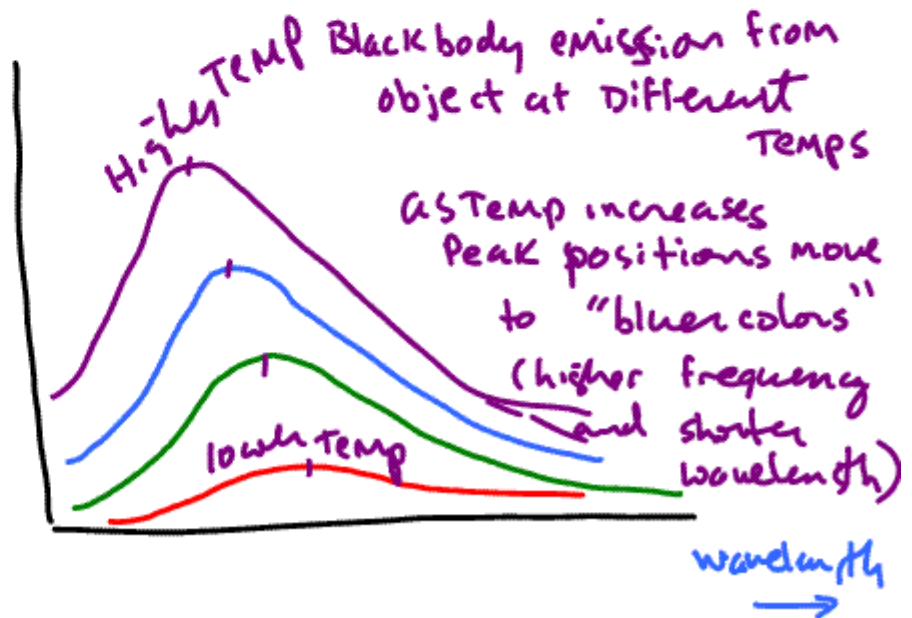
<http://www-history.mcs.st-andrews.ac.uk/Mathematicians/Planck.html>



observer  
Light intensity



"Blackbody"  
radiation  
(emitted by object)



AT time w/ best models of atoms and understanding of light as a wave physicists could NOT understand blackbody spectra theoretically.

Planck Succeeded!

But to do so he hypothesized

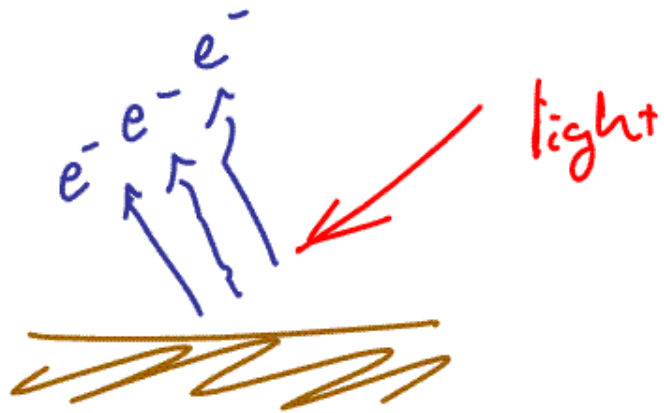
light  $\rightarrow$  little packets w/ Energy given by

$$E = h \nu$$

Energy  $\nearrow$   $\uparrow$   $\uparrow$  freq.  $\Rightarrow$  Blackbody  
CONSTANT  
Planck's constant

Planck's Theory worked perfectly ... but physicists thought it was a fortuitous accident ... after all, light is a wave.

Another phenomenon that was NOT understood



## Photoelectric effect

Study electric current  
depends on Intensity  
+ color of light

Einstein  
in 1905

$$\underline{\underline{E = h\nu}}$$

Einstein able to explain photoelectric effect  
Assuming light to come in little packets  
with energy  $E = h\nu$

Suddenly world has to face fact that light  
is both a wave and a particle  
... Sounds CRAZY ... but that's what nature tells us.

It gets worse ... if light (known to be a wave) can be a particle, perhaps particles like electrons can be waves



Prince Louis-Victor Pierre  
Raymond de Broglie  
1892 - 1987 (France)

1929 Nobel Prize in physics

(1924 - doctoral thesis)

Matter  
Waves

$$\lambda = \frac{h}{p}$$

recall  $p \sim mv$

plank's constant

wavelength of particle of matter

Momentum

Yikes!

quantum mechanics  
is born